

LITERATURE REPORT

Nickel-Catalyzed Reductive Cyclizations and Couplings of Alkyne and Aldehyde

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Supervisor: Prof. Yong Huang

2015.05.25

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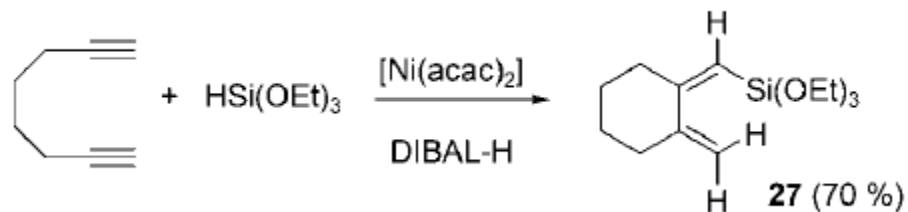
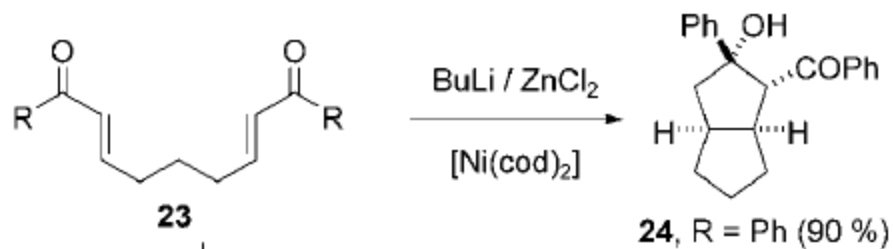
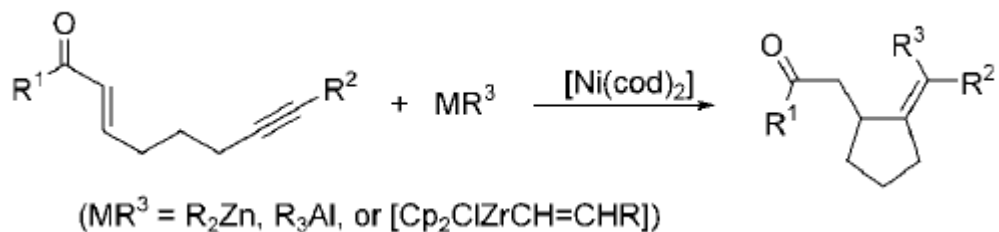
5. Summary

J. Montgomery, *et al. Acc. Chem. Res.* **2015**, *48*, ASAP.

T. F. Jamison, *et al. Acc. Chem. Res.* **2015**, *48*, ASAP.

S. Ogoshi, *et al. Acc. Chem. Res.* **2015**, *48*, ASAP.

1. Introduction:

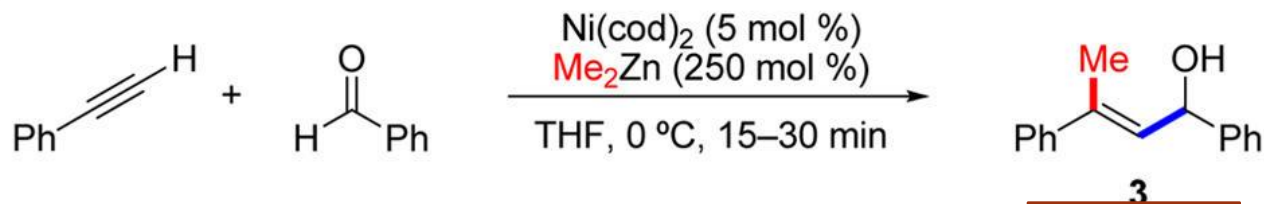


J. Montgomery, *et al. Angew. Chem., Int. Ed.* **2004**, 43, 3890.

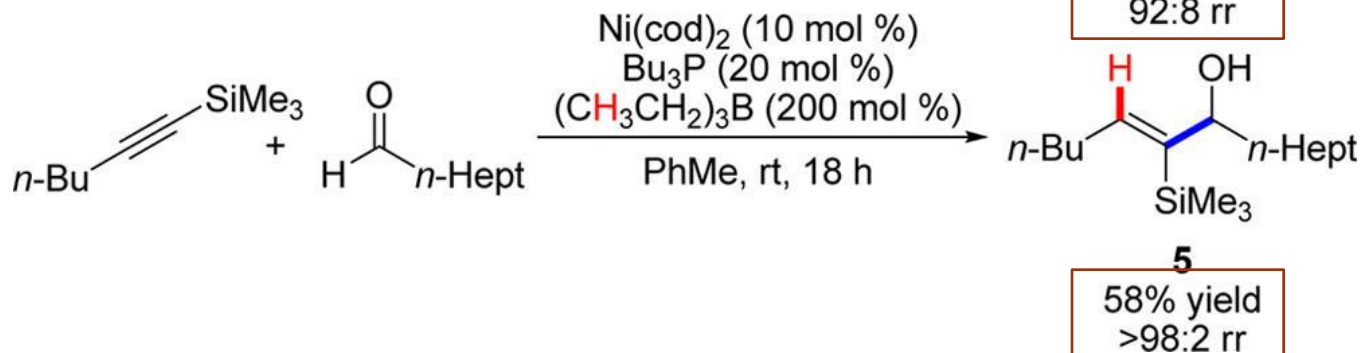
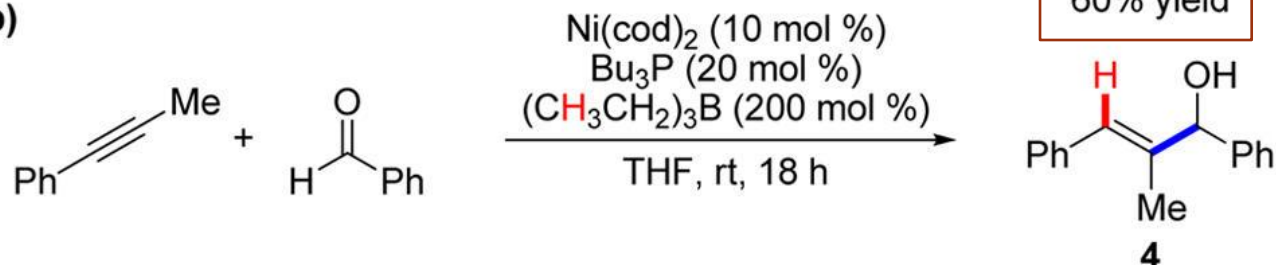
2. Reductive Coupling of Alkyne and Aldehyde

Early studies:

a)



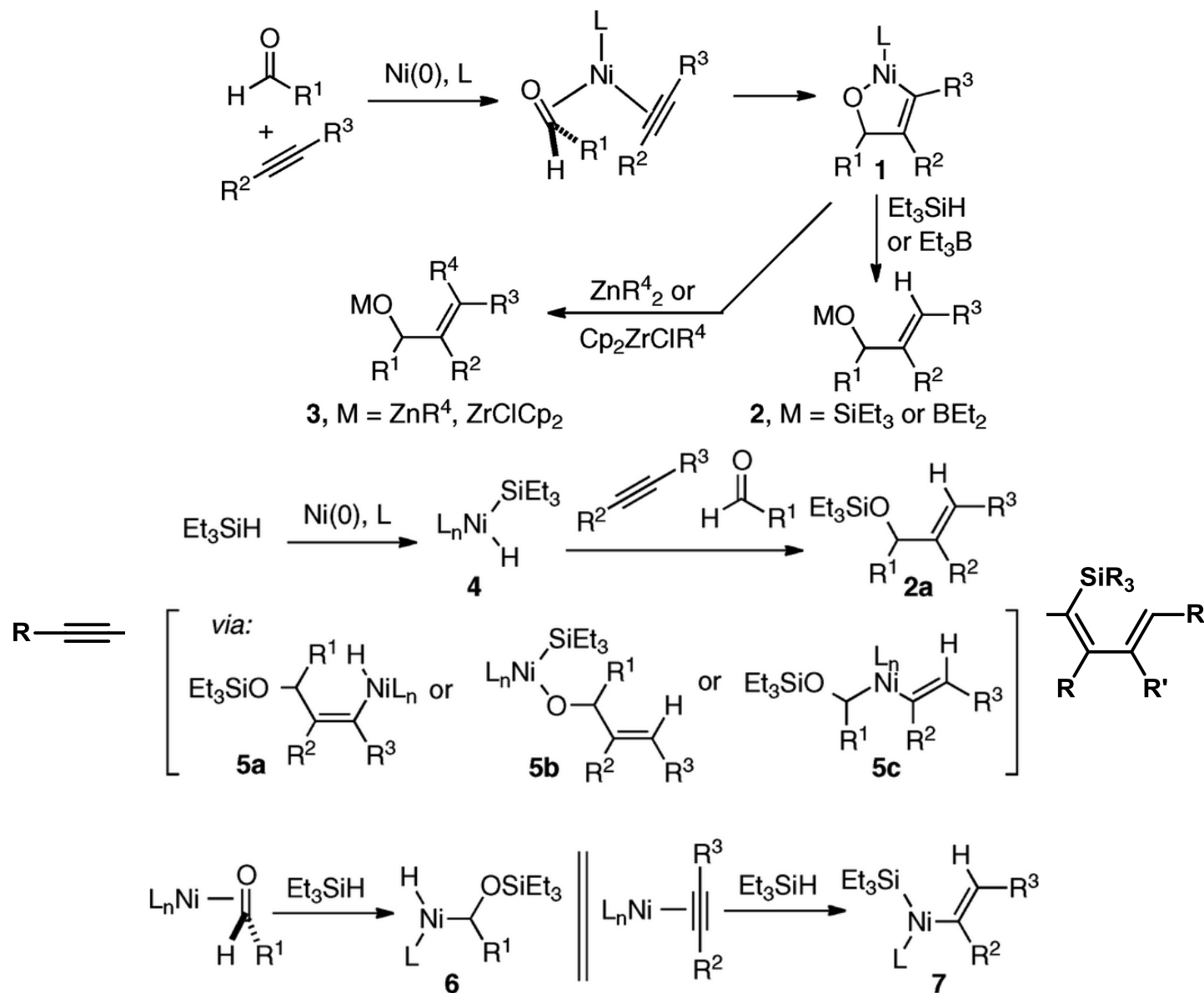
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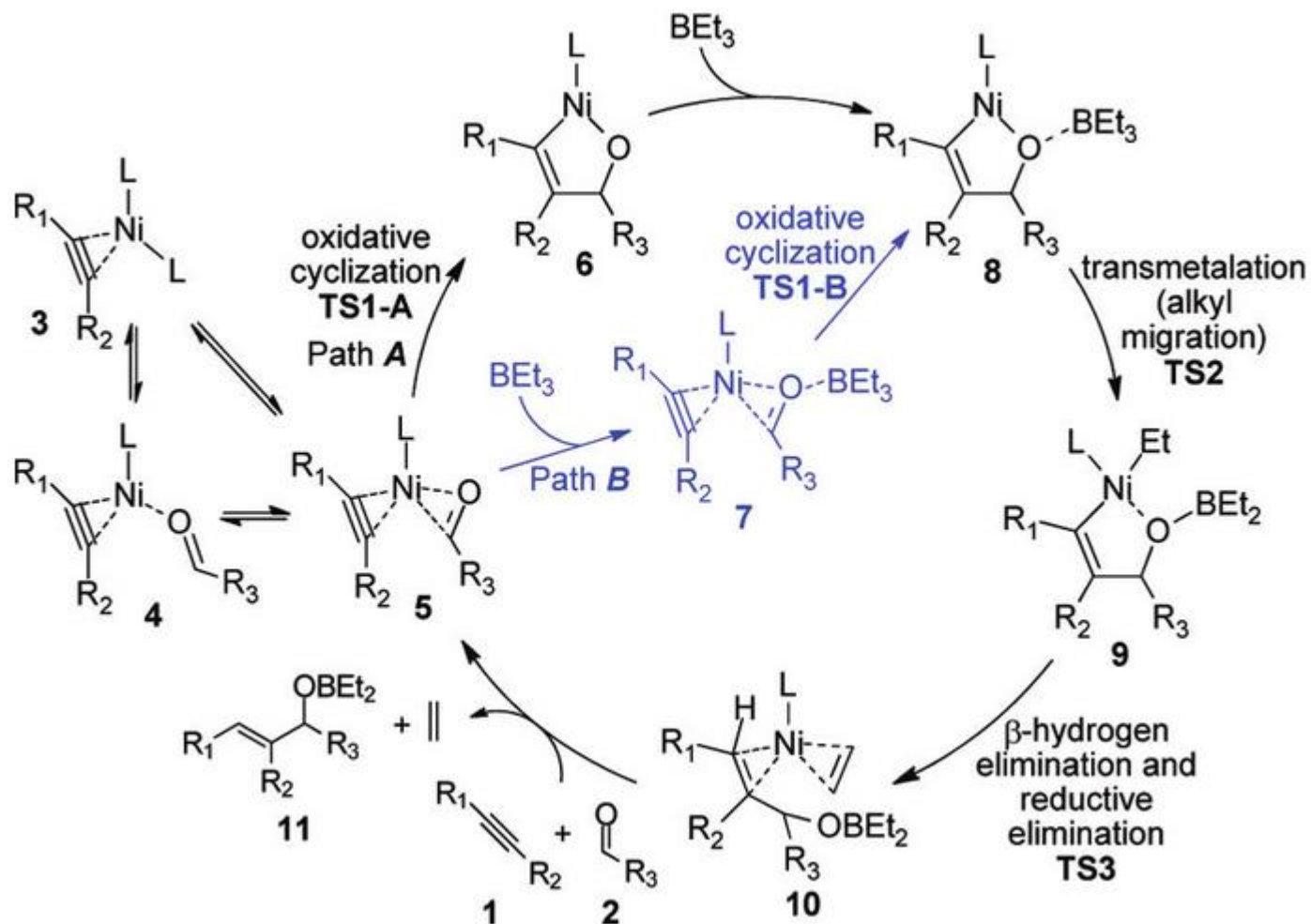
J. Montgomery, *et al.* *J. Am. Chem. Soc.* **1997**, *119*, 9065.

T. F. Jamison, *et al.* *Org. Lett.* **2000**, *2*, 4221.

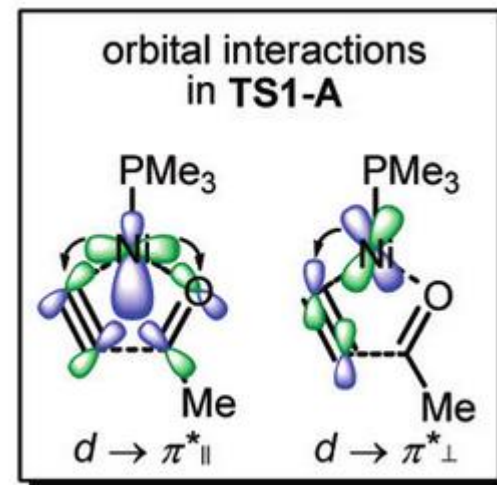
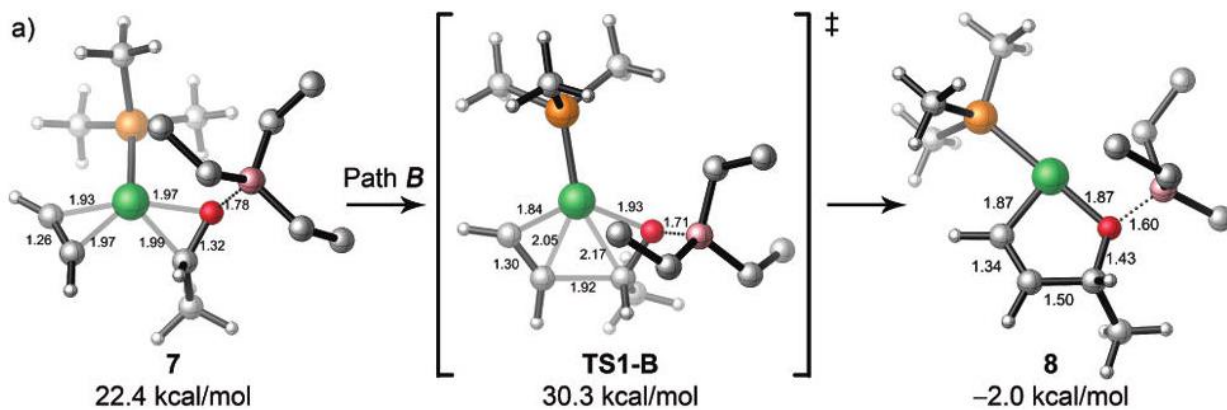
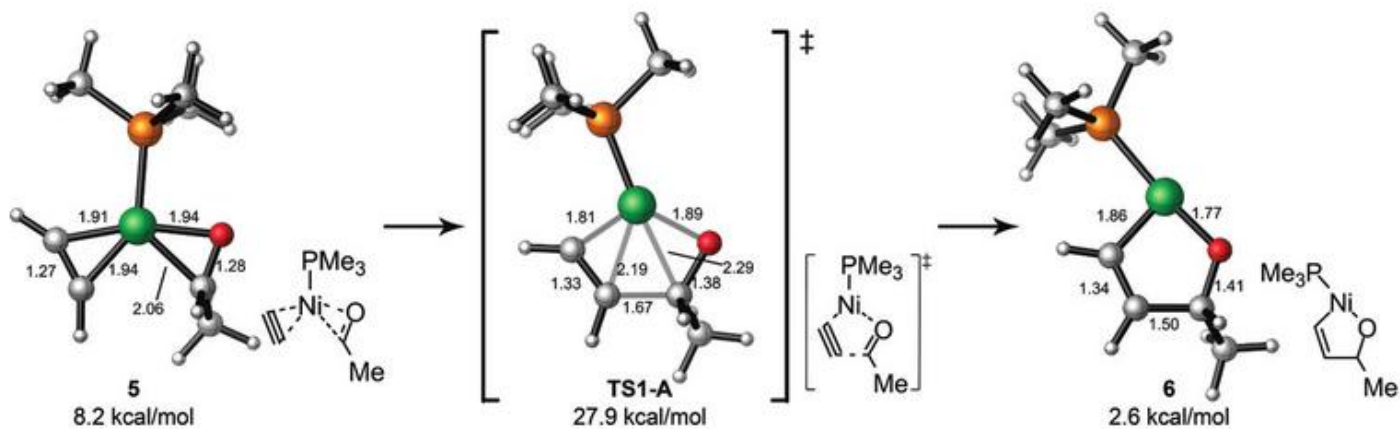
Mechanism discussion:



Computational study:

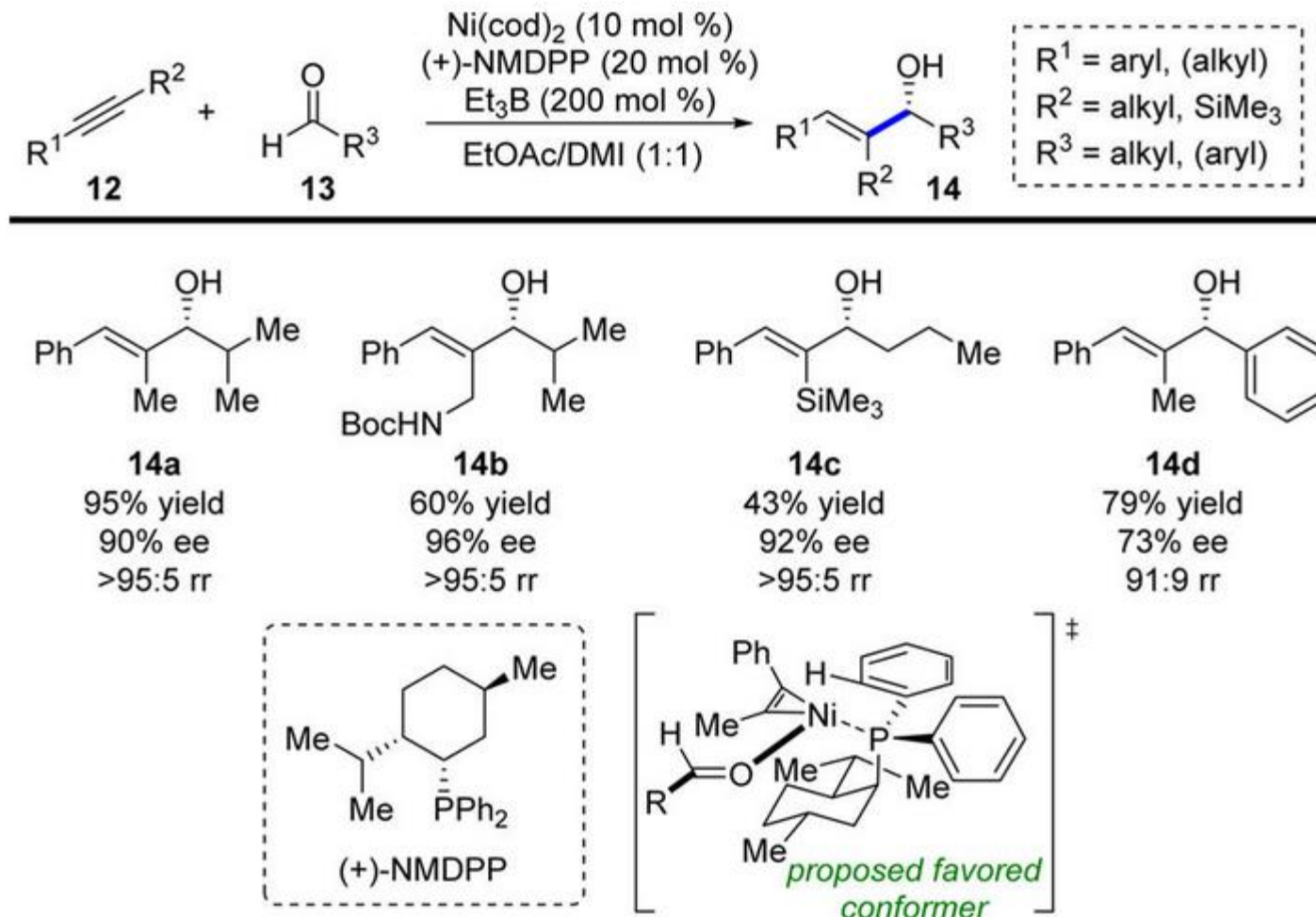


K. N. Houk and T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2009**, *131*, 6654.



K. N. Houk and T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2009**, 131, 6654.

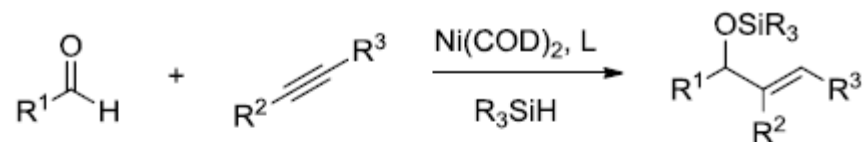
Enantioselective alkyne-aldehyde reductive coupling:



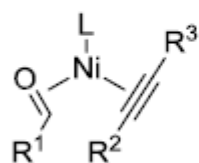
T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2003**, 125, 3442.

T. F. Jamison, *et al.* *J. Org. Chem.* **2003**, 68, 156.

Regiocontrol strategies:



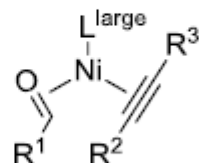
Electronic Control



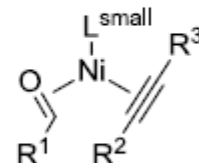
favored by:

$R^2 = \text{H}, \text{SiR}_3, \text{CH}_2\text{OR}$
 $R^3 = \text{aryl, alkenyl, NR}_2$

Steric Control

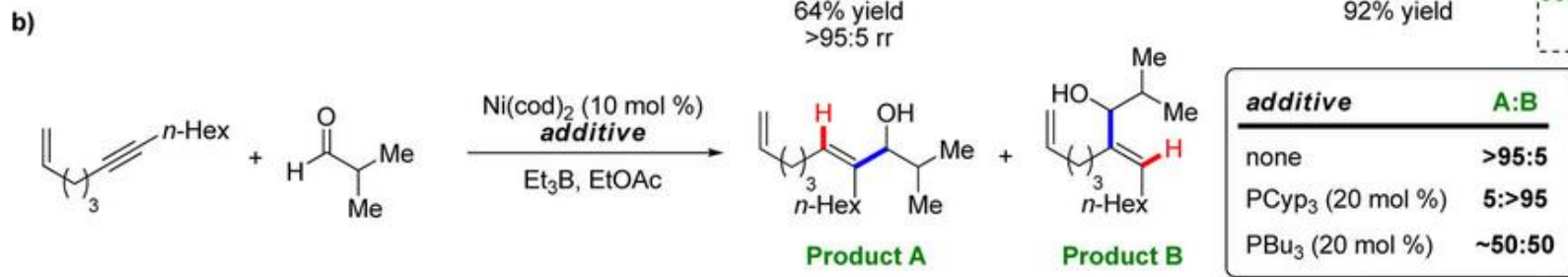
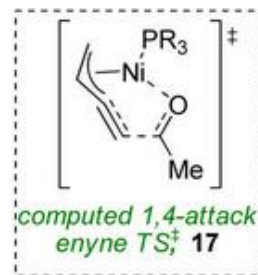
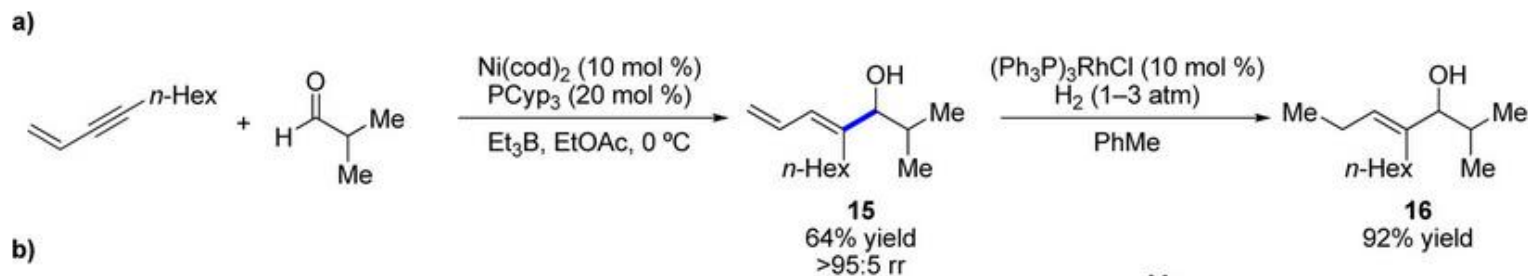


*where R^2 is
larger than R^3*



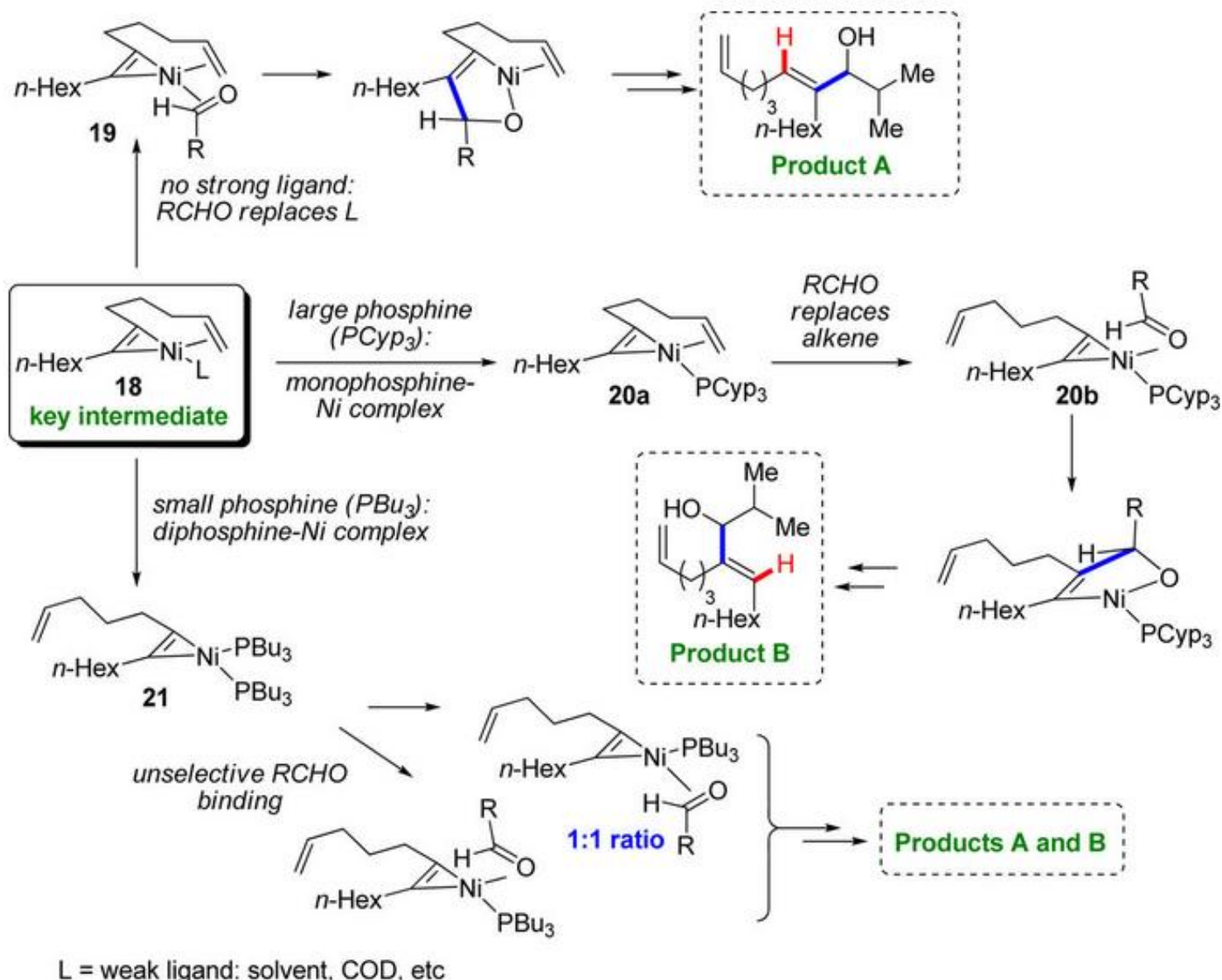
*where R^2 is
smaller than R^3*

Alkene directing effects:



reference alkynes (not alkene-directed)	alkene-directed (this work)	effect of alkenyl group
>20 1	1a 1 >20	reverses regioselectivity
(do not couple)	1b 1 >20	increases reactivity – and – controls regioselectivity
2 1	1d 1 >20 1f >20 1	circumvents poor regioselectivity

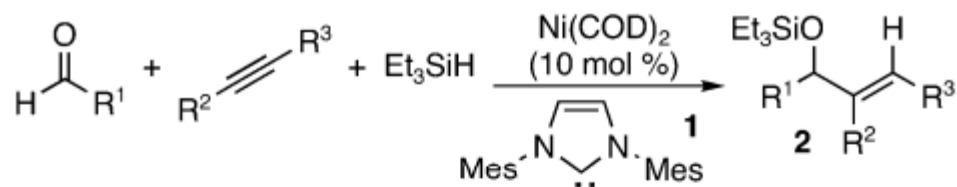
Alkene directing effects: mechanism study



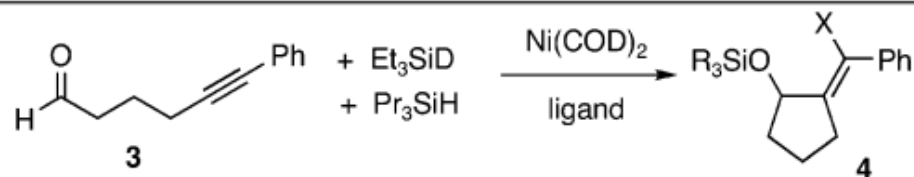
T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2004**, 126, 4130.

K. N. Houk and T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2010**, 132, 2050.

NHCs ligand effects:

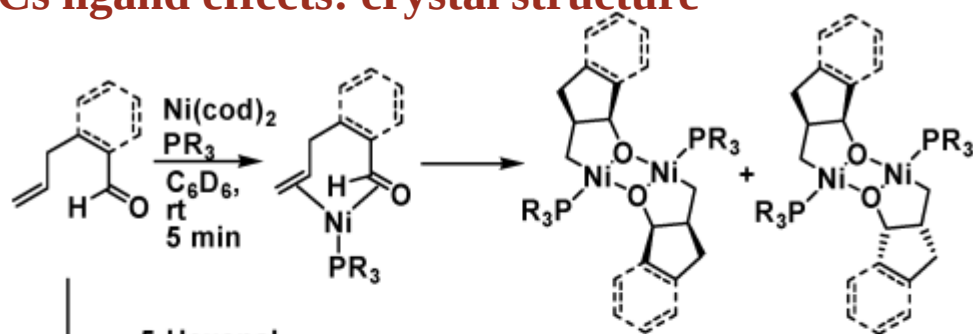


entry	R^1	R^2	R^3	yield (regioselectivity)
1	Ph	CH_3	Ph	84% (>98:2)
2	C_6H_{13}	CH_3	Ph	82% (>98:2)
3	Ph	H	C_6H_{13}	71% (>98:2)
4	Ph	H	Ph	72% (>98:2)
5	Ph	CH_3	C_4H_9	84% (1.3:1)
6	<i>s</i> -Bu	CH_3	Ph	81% (>98:2) ^c
7	$\text{C}_6\text{H}_4\text{OCH}_3$	CH_3	Ph	66% (>98:2)
8	Ph	Ph	$\text{C}(\text{CH}_3)=\text{CH}_2$	84% (>98:2)
9	Ph	H	$\text{CH}=\text{CHC}_6\text{H}_{13}$	56% (>98:2)
10	Ph	H	$(\text{CH}_2)_4\text{OH}$	72% (>98:2) ^d



R	X	product	relative %	
			from 1	from PBU_3
Et	H	4a	<2	25
Et	D	4b	55	34
Pr	H	4c	41	23
Pr	D	4d	<2	18

NHCs ligand effects: crystal structure



5-Hexenal

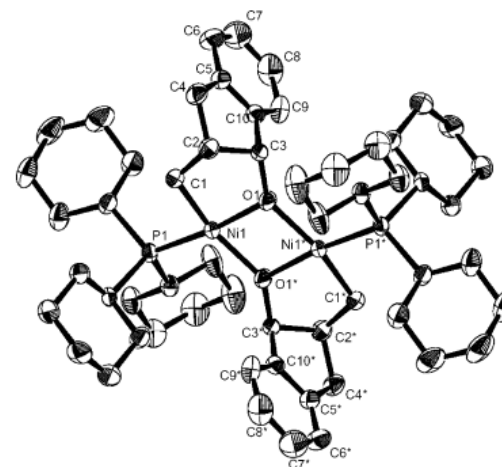
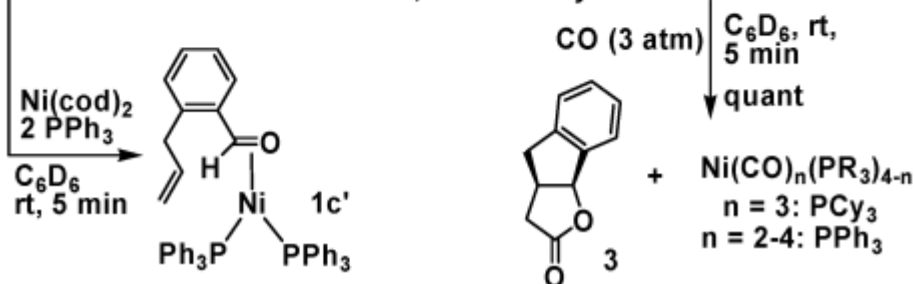
PR₃ = PCy₃ 1a 80 °C, 14 h 2a-syn: 56%

o-Allylbenzaldehyde

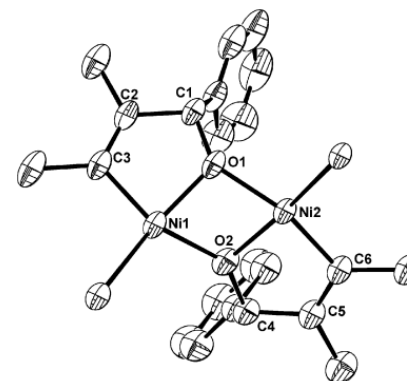
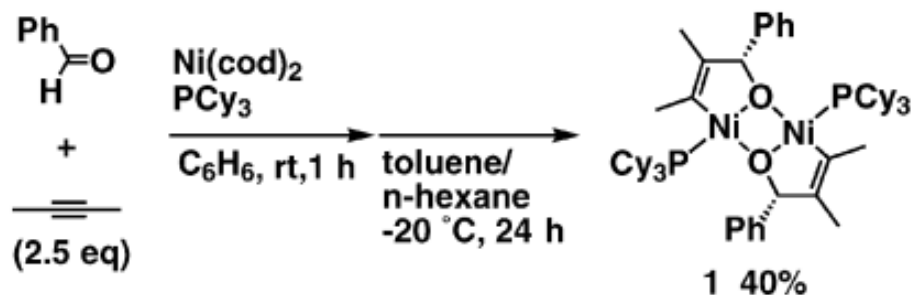
PR₃ = PCy₃ 1b 60 °C, 24 h 2b-syn: 68% 2b-anti: 30%

PPh₃ 1c 60 °C, 8 h 2c-syn: 57% 2c-anti: 43%

1c 25 °C, 16 h 2c-syn: 36% 2c-anti: 31%

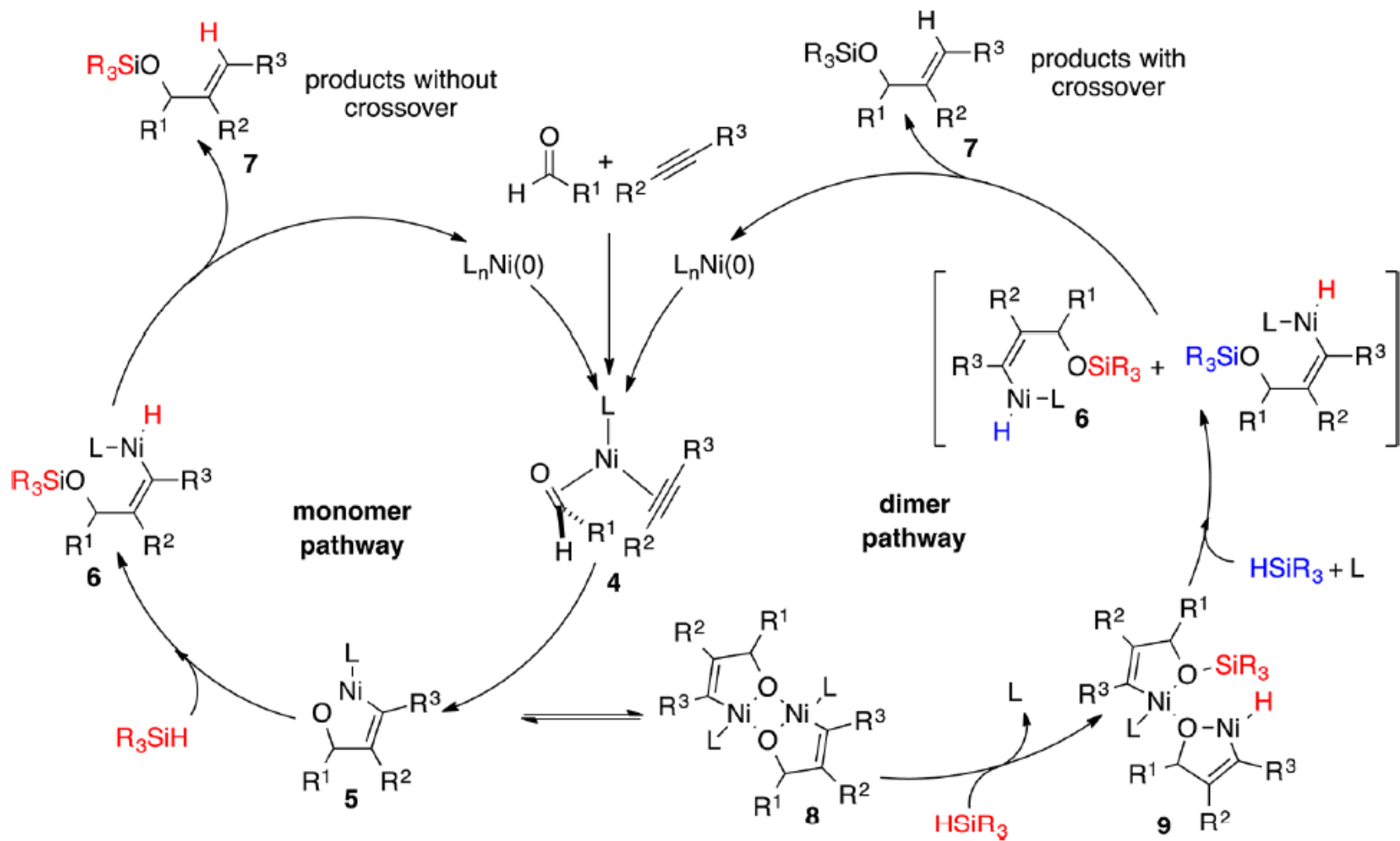


S. Ogoshi, *et al.* *J. Am. Chem. Soc.* **2004**, 126, 11802.

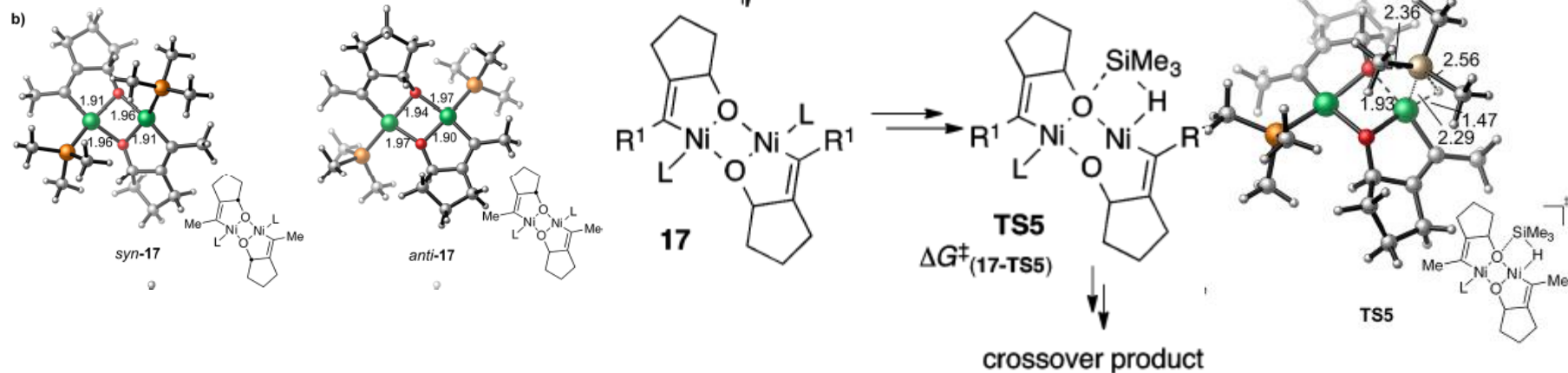
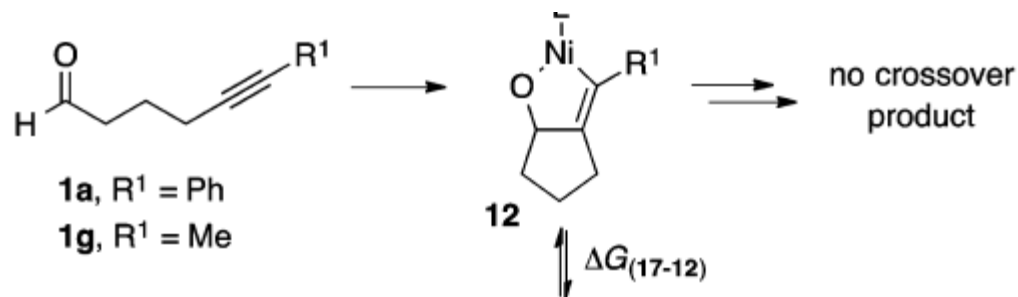


S. Ogoshi, *et al.* *Chem. Comm.* **2008**, 44, 1347.

NHCs ligand effects: computational study



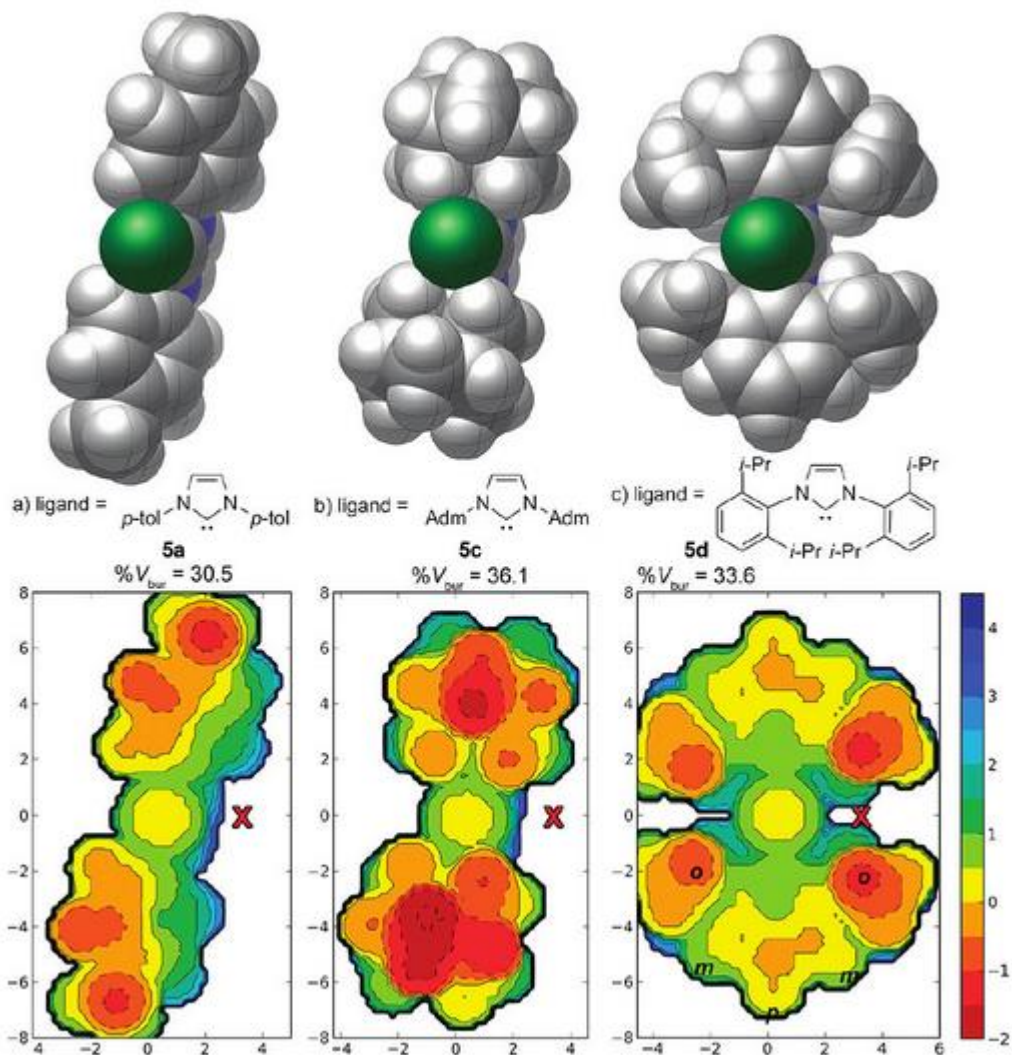
K. N. Houk and J. Montgomery, *et al.* *J. Am. Chem. Soc.* **2014**, 136, 17495.



entry	ligand	ynal	$\Delta G_{17 \rightarrow 12}$	$\Delta G^\ddagger_{17 \rightarrow \text{TS5}}$
1	PMe_3	1g	20.3	25.9
2	$\text{P}(i\text{-Pr})_3$	1g	15.2	24.2
3	IPh	1g	23.7	44.9
4	IMes	1g	14.0	36.1
5	PMe_3	1a	22.3	31.8

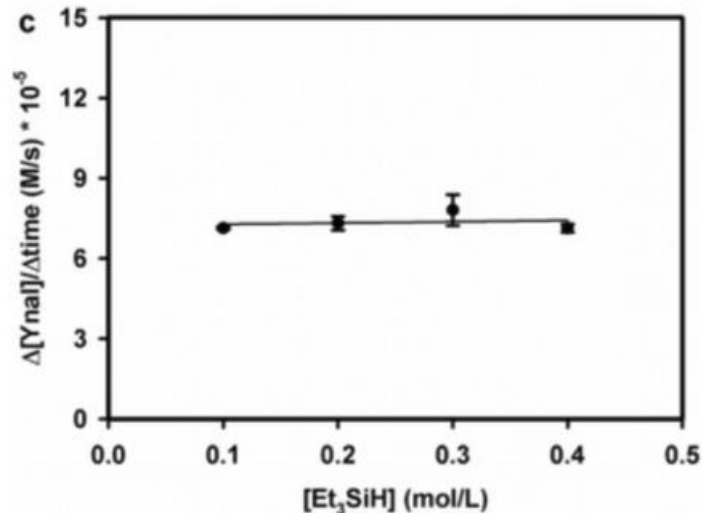
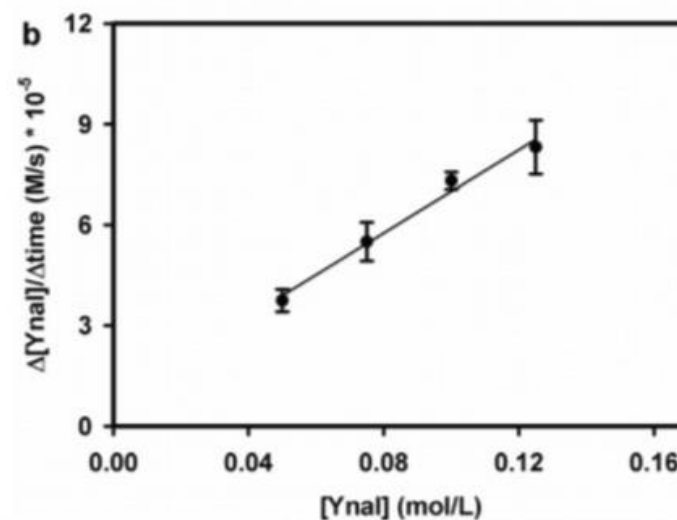
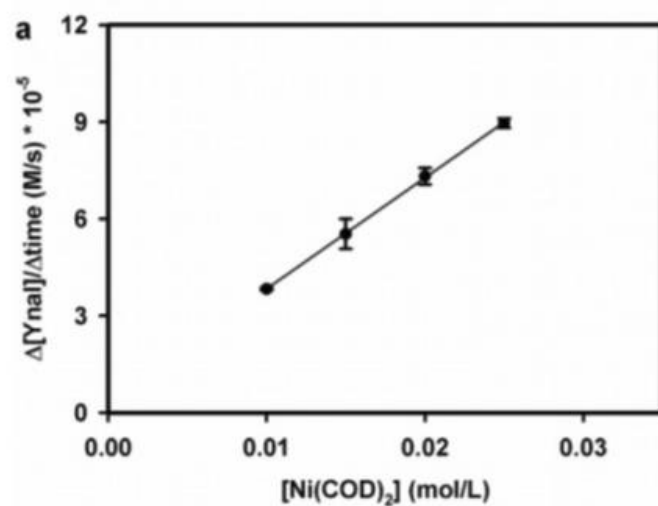
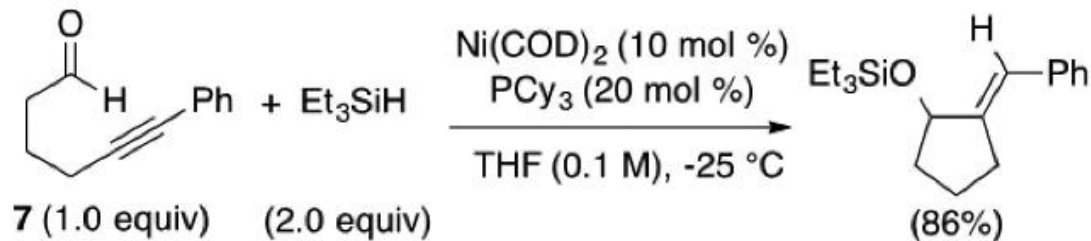
^aEnergies are in kilocalories per mole and with respect to **17**.

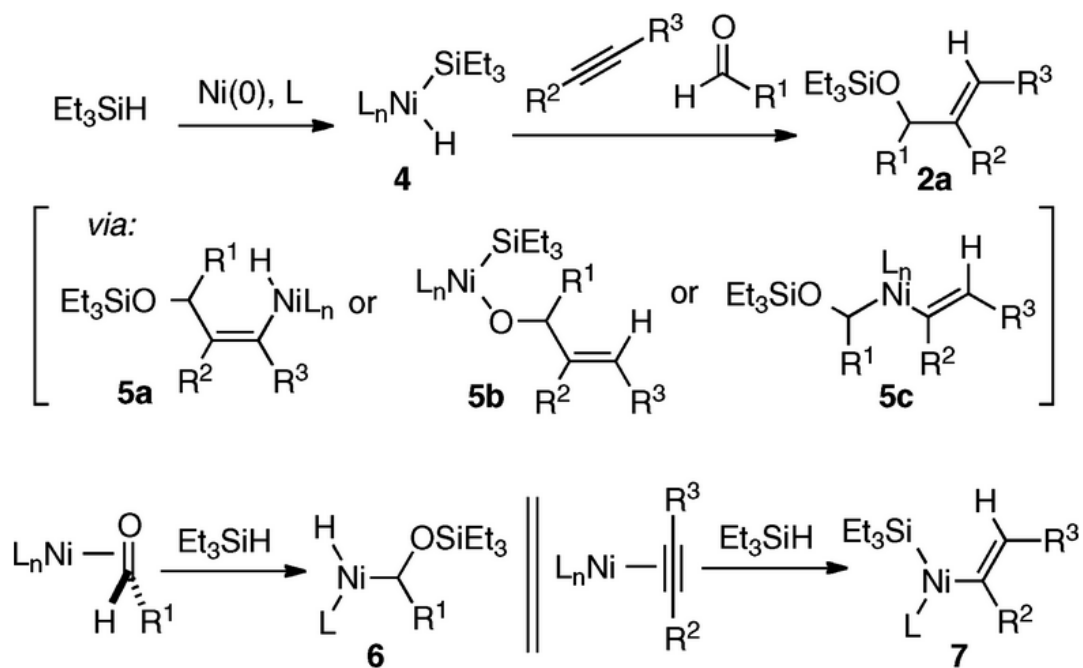
K. N. Houk and J. Montgomery, *et al.* *J. Am. Chem. Soc.* **2014**, 136, 17495.



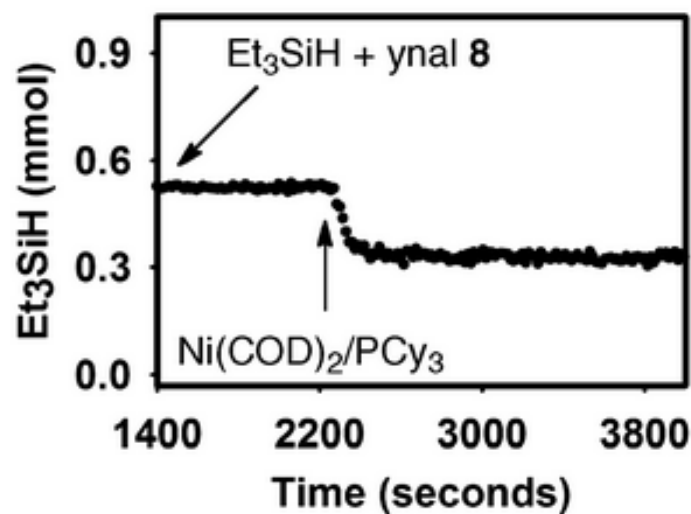
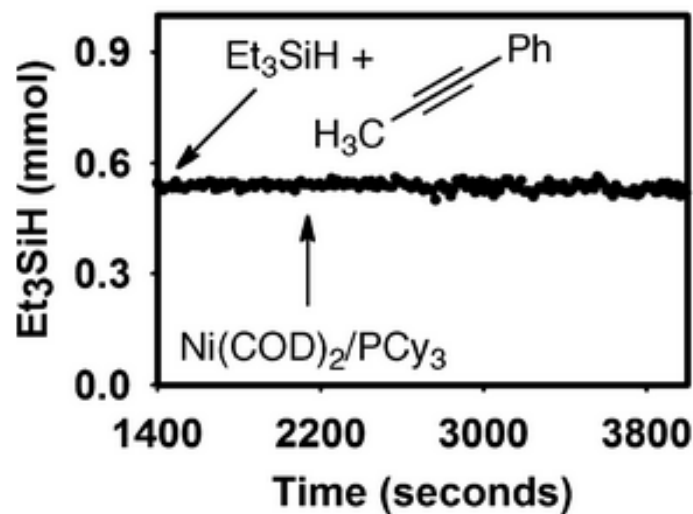
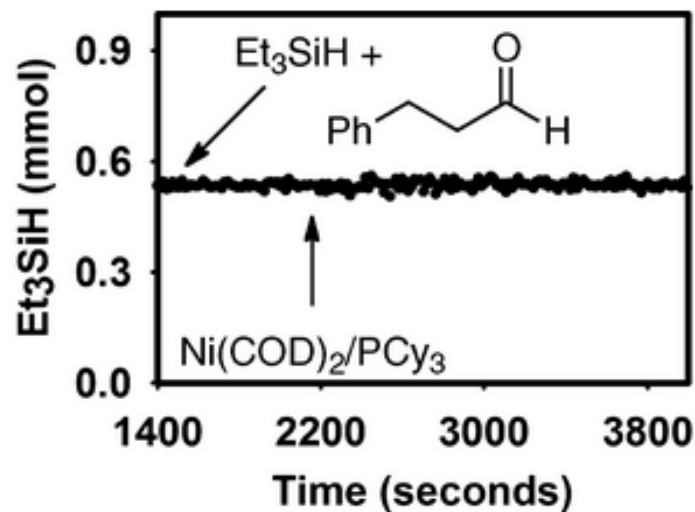
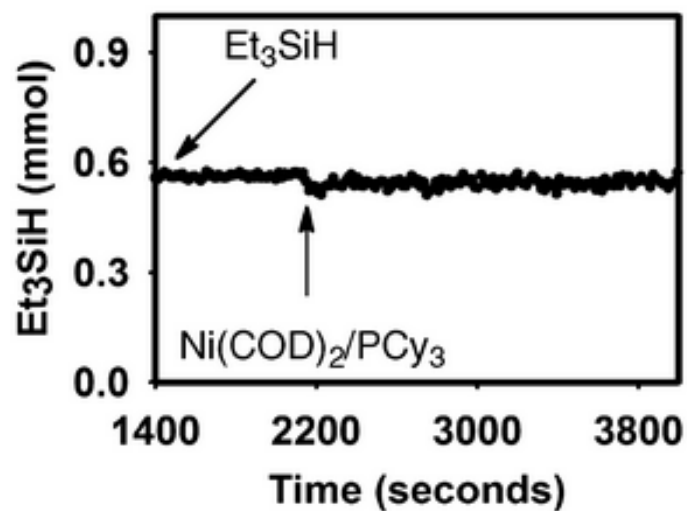
K. N. Houk and J. Montgomery, *et al.* *J. Am. Chem. Soc.* **2011**, 133, 6956.

Kinetic study:

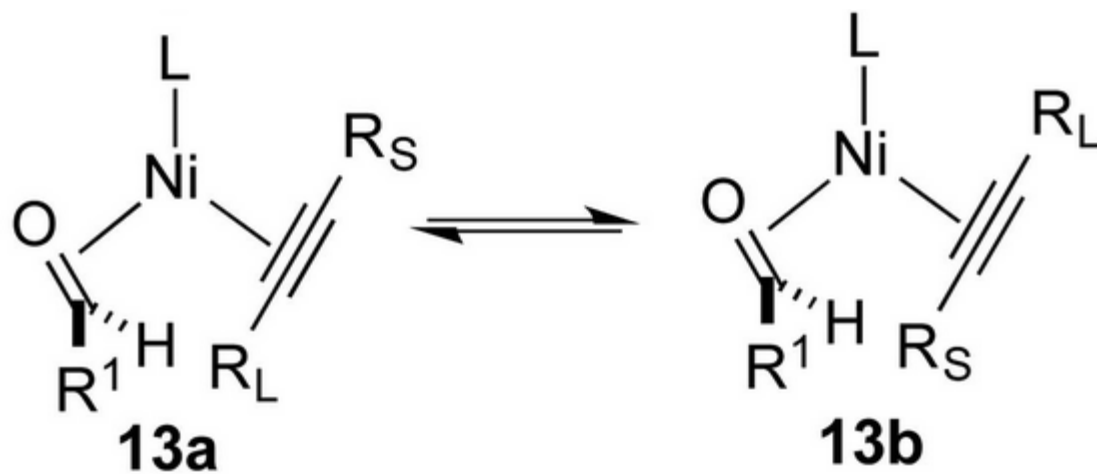




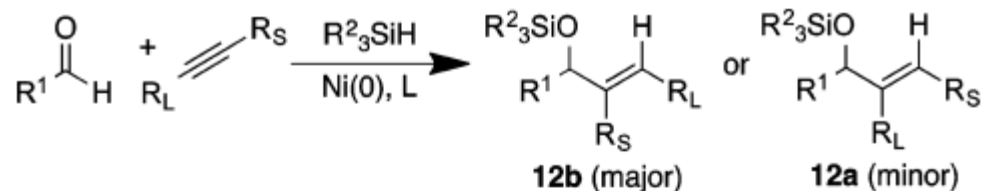
J. Montgomery, *et al.* *J. Am. Chem. Soc.* **2011**, 133, 5728.



A general strategy for regiocontrol:

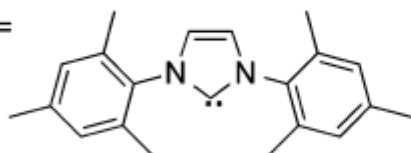


A general strategy for regiocontrol: small ligand protocol

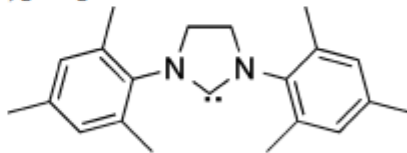


entry	R ¹	R ^S	R ^L	R ² ₃ SiH	L	% yield	12b/12a
1	<i>n</i> -Hex	H	<i>i</i> -Pr	(<i>i</i> -Pr) ₃ SiH	16	74	>98:2
2	Ph	Me	Ph	(<i>i</i> -Pr) ₃ SiH	16	74	>98:2
3	<i>n</i> -Hex	Me	<i>c</i> -hexenyl	(<i>i</i> -Pr) ₃ SiH	16	99	97:3
4	<i>n</i> -Hex	Me	<i>n</i> -Pr	(<i>i</i> -Pr) ₃ SiH	16	83	67:33
5	<i>n</i> -Hex	Me	<i>n</i> -Pr	(<i>i</i> -Pr) ₃ SiH	17	73	61:39
6	<i>n</i> -Hex	Me	<i>n</i> -Pr	(<i>i</i> -Pr) ₃ SiH	18	18	87:13
7	<i>n</i> -Hex	Me	<i>n</i> -Pr	(<i>t</i> -Bu) ₂ SiH ₂	19	78	88:12

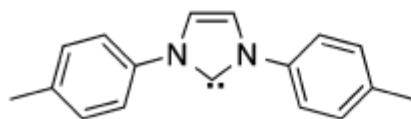
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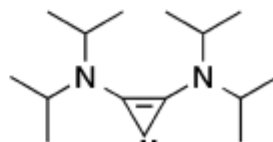
IMes (16)



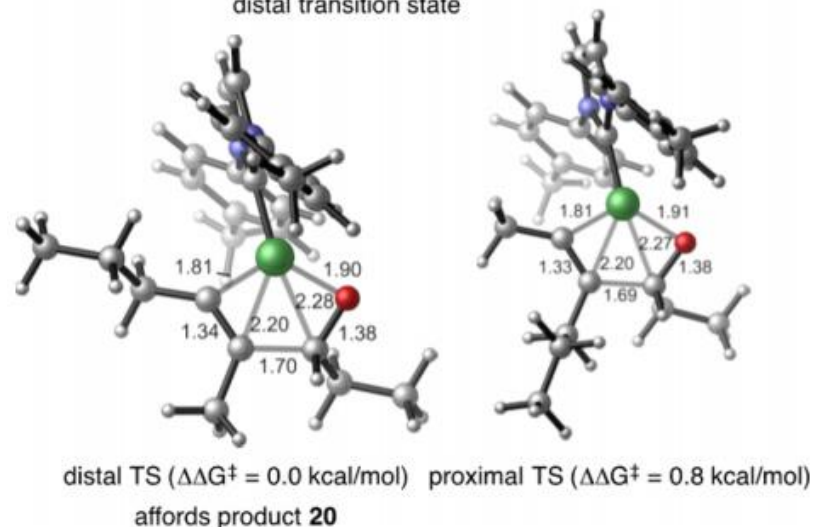
sIMes (17)



ITol (18)



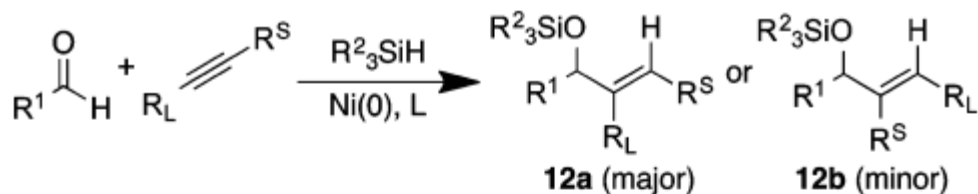
IPr-BAC (19)



J. Montgomery, *et al.* *J. Am. Chem. Soc.* **2010**, 132, 6304.

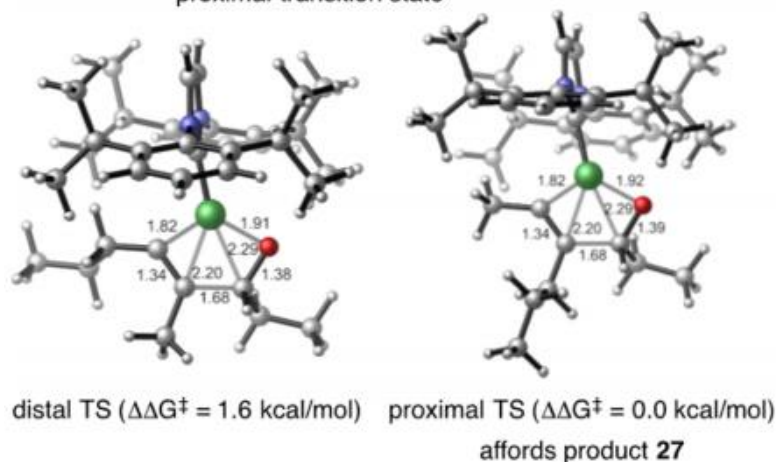
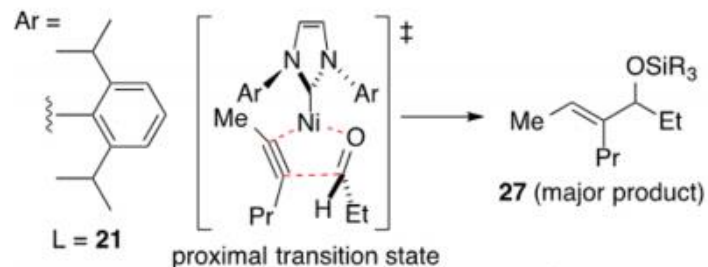
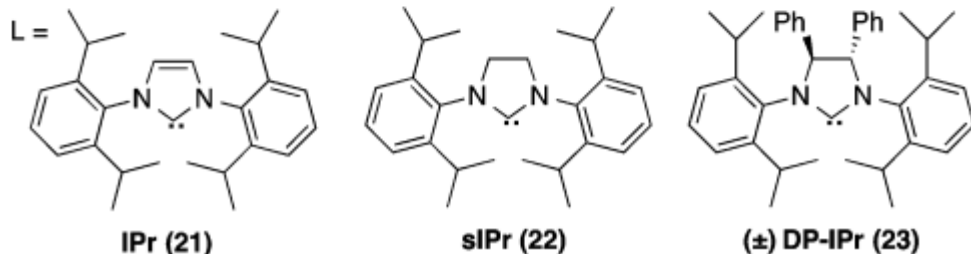
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A general strategy for regiocontrol: large ligand protocol



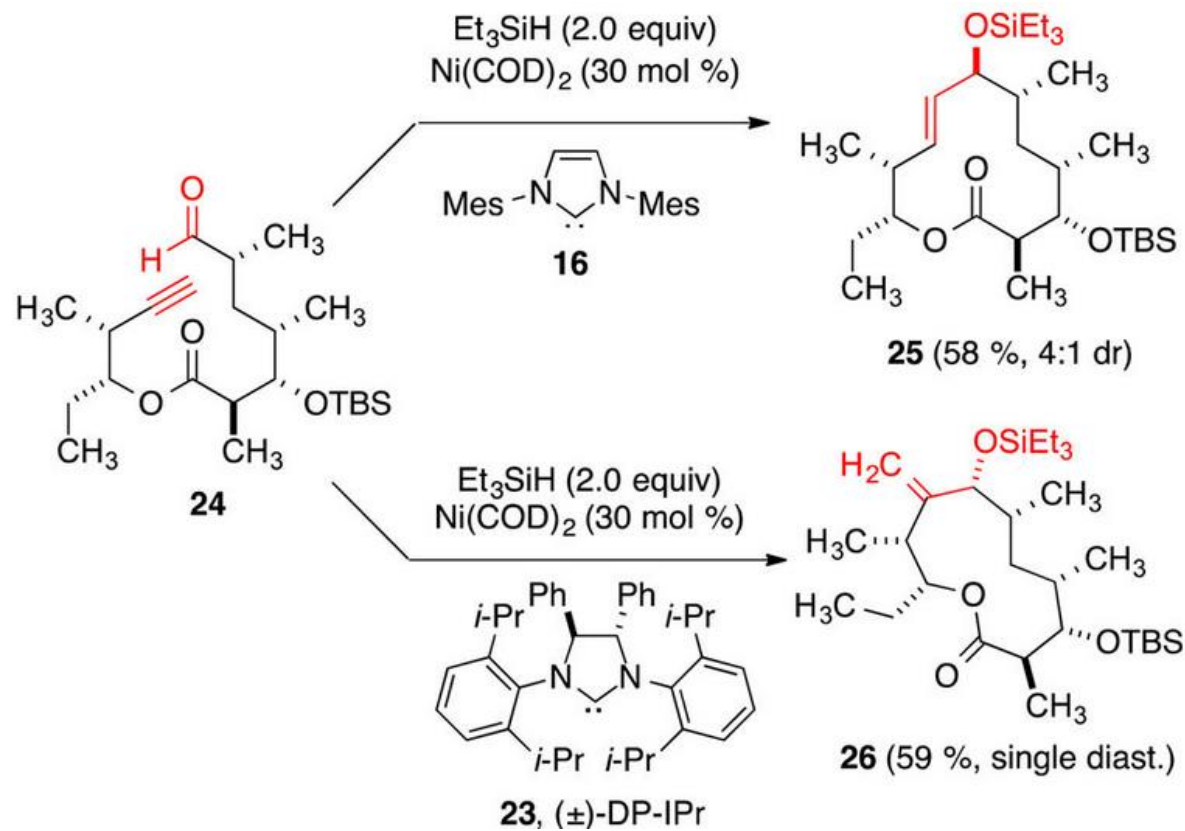
entry	R ¹	R ^L	R ^S	R ² ₃ SiH	L	% yield	12a/12b
1	<i>n</i> -Hex	<i>n</i> -Pr	Me	(<i>i</i> -Pr) ₃ SiH	16	83	67:33
2	<i>n</i> -Hex	<i>n</i> -Pr	Me	(<i>i</i> -Pr) ₃ SiH	21	84	80:20
3	<i>n</i> -Hex	<i>n</i> -Pr	Me	(<i>i</i> -Pr) ₃ SiH	22	85	93:7
4	<i>n</i> -Hex	<i>n</i> -Pr	Me	Et ₃ SiH	23	94	94:6
5	<i>n</i> -Hex	<i>i</i> -Pr	H	(<i>i</i> -Pr) ₃ SiH	22	40	41:59 ^a
6	<i>n</i> -Hex	<i>i</i> -Pr	H	Et ₃ SiH	23	76	95:5
7	Ph	Ph	Me	Et ₃ SiH	22	65	58:42
8	Ph	Ph	Me	Et ₃ SiH	23	96	69:31 ^a

^aData from H. A. Malik thesis, University of Michigan.

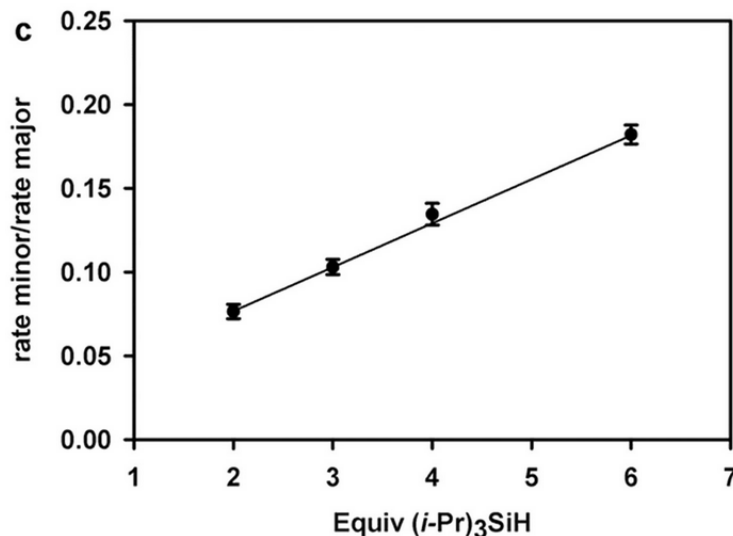
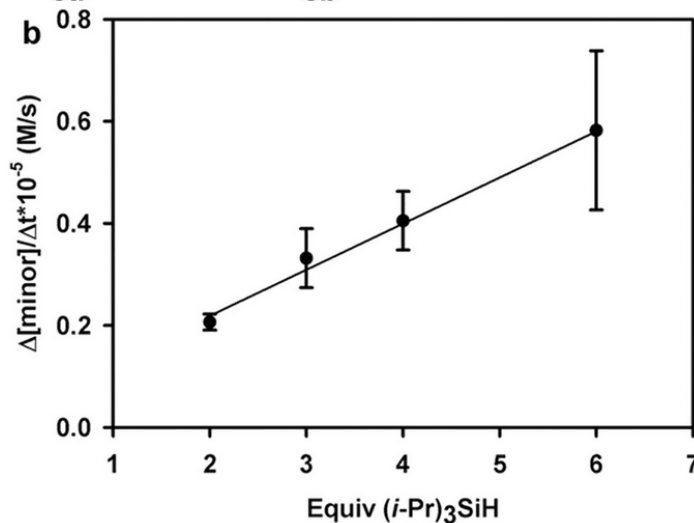
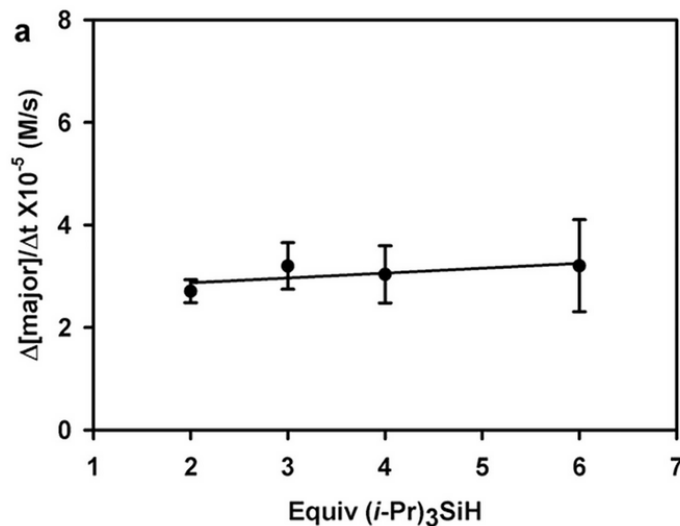
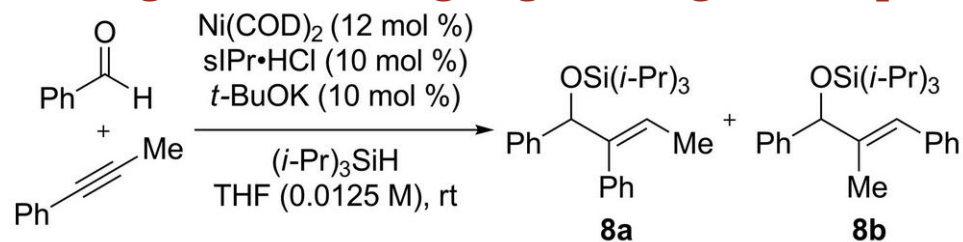


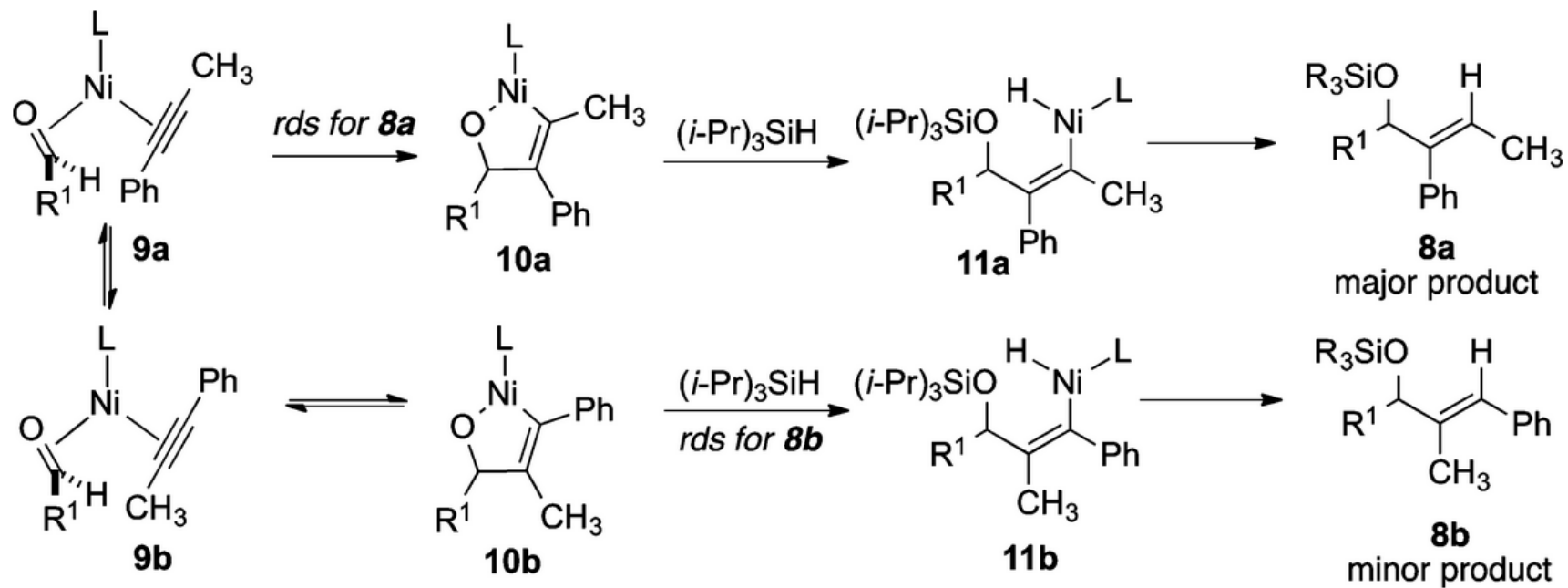
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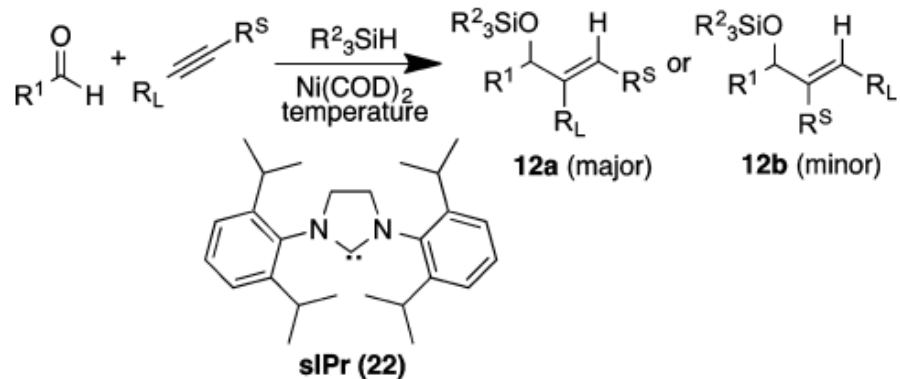


A general strategy for regiocontrol: large ligand/large silane protocol





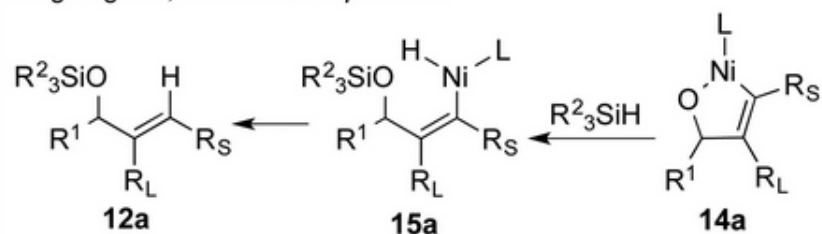
J. Montgomery, et al. *J. Am. Chem. Soc.* **2015**, 137, 958.



entry	R ¹	R ^L	R ^S	R ² ₃ SiH	temp (°C)	% yield	12a/12b
1	Ph	Ph	Me	(<i>i</i> -Pr) ₃ SiH	25	82	>98:2
2	<i>n</i> -Hept	Ph	Me	(<i>i</i> -Pr) ₃ SiH	50	77	>98:2
3	<i>n</i> -Hept	<i>i</i> -Bu	Et	(<i>i</i> -Pr) ₃ SiH	50	66	93:7
4	Ph	<i>i</i> -Pr	Me	(<i>i</i> -Pr) ₃ SiH	50	78	>98:2
5	Ph	<i>n</i> -Pr	Et	(<i>i</i> -Pr) ₃ SiH	50	56	68:32
6	<i>n</i> -Hex	<i>c</i> -hexenyl	Me	(<i>i</i> -Pr) ₃ SiH	25	77	91:9
7	Ph	<i>i</i> -Pr	H	(<i>t</i> -Bu) ₂ MeSiH	25	61	>98:2
8	Ph	<i>n</i> -Hex	H	(<i>t</i> -Bu) ₂ MeSiH	25	69	95:5

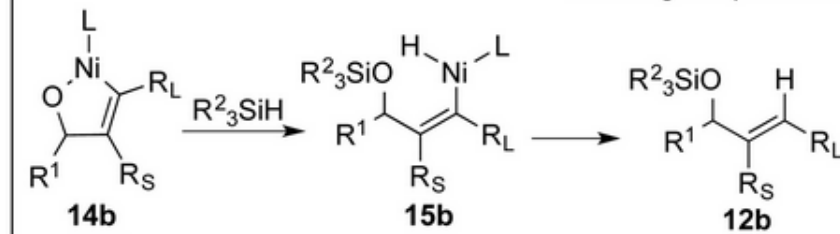
J. Montgomery, *et al.* *J. Am. Chem. Soc.* **2015**, *137*, 958.

Large ligand, small silane protocol

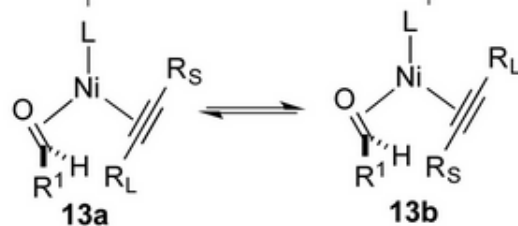


Governed by rate of formation of **14a** vs **14b**

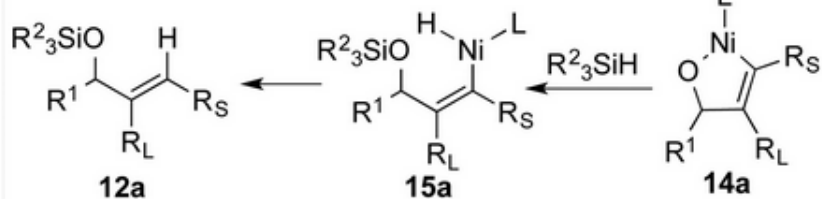
Small ligand protocol



Governed by rate of formation of **14b** vs **14a**



Large ligand, large silane protocol

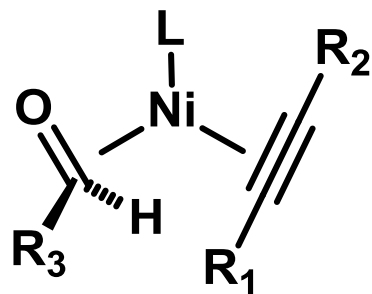


Governed by rate of formation of **14a** vs **15b**

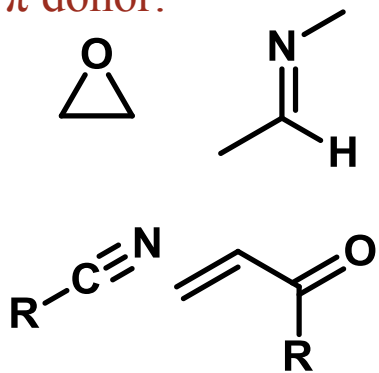


J. Montgomery, *et al. Acc. Chem. Res.* **2015**, *48*, ASAP.

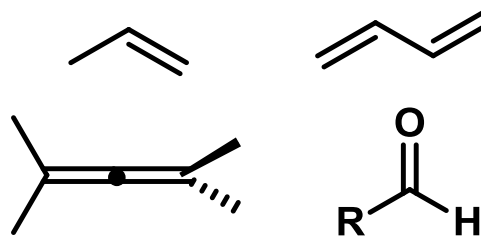
3. Reductive Coupling of Different reaction partner



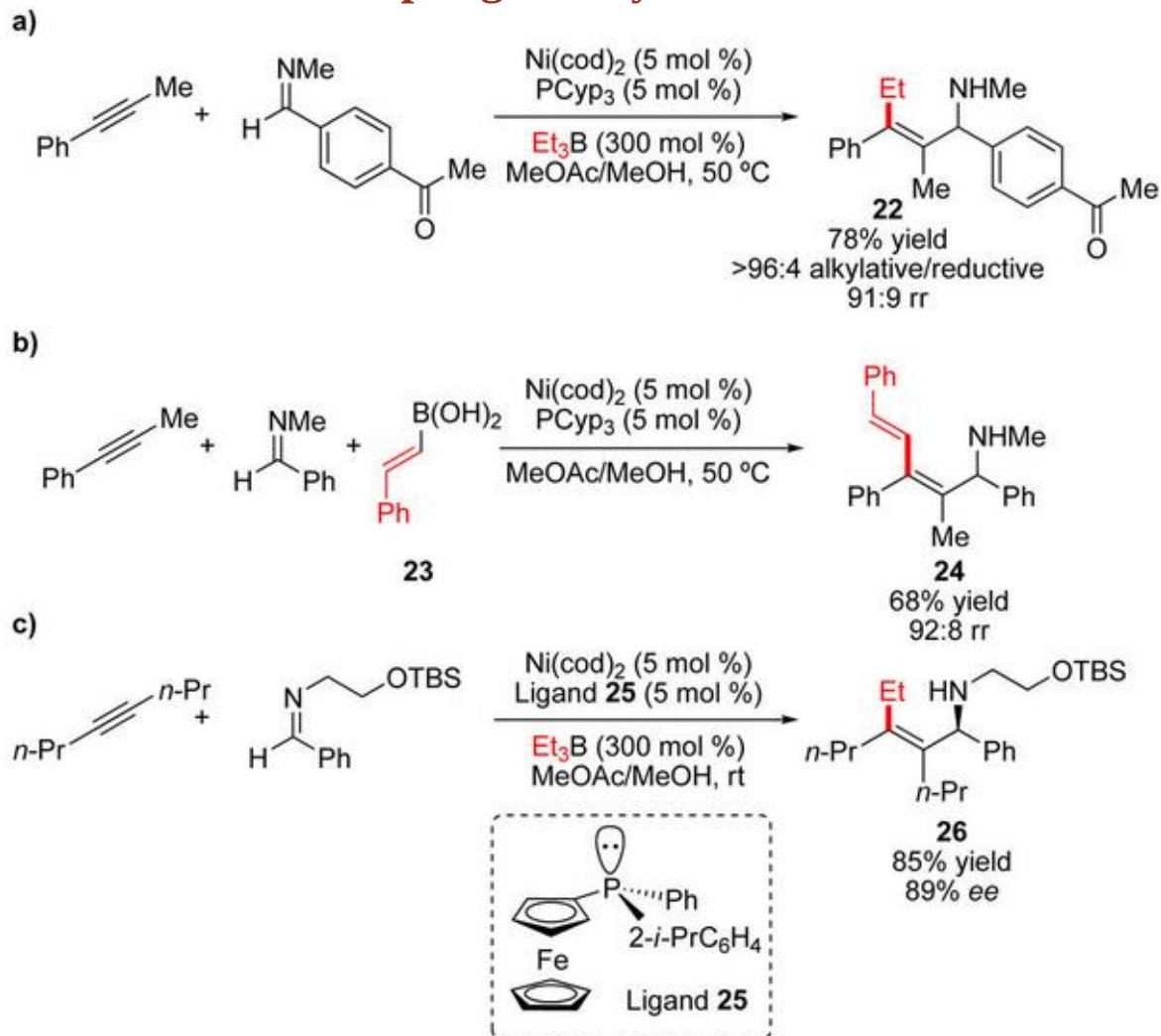
π donor:



π component:

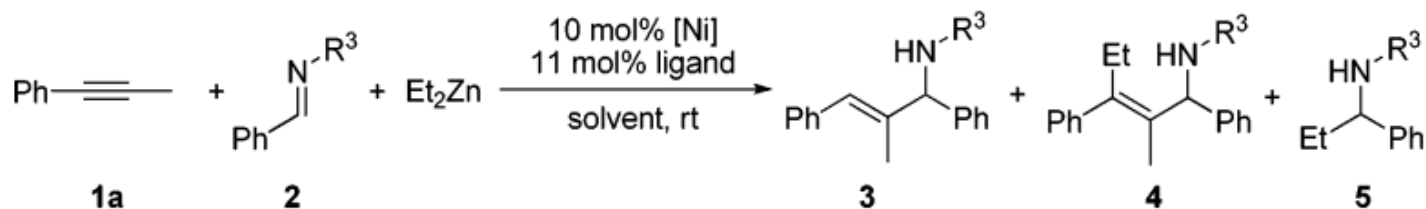


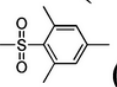
Selected examples of reductive coupling of alkyne and imines:

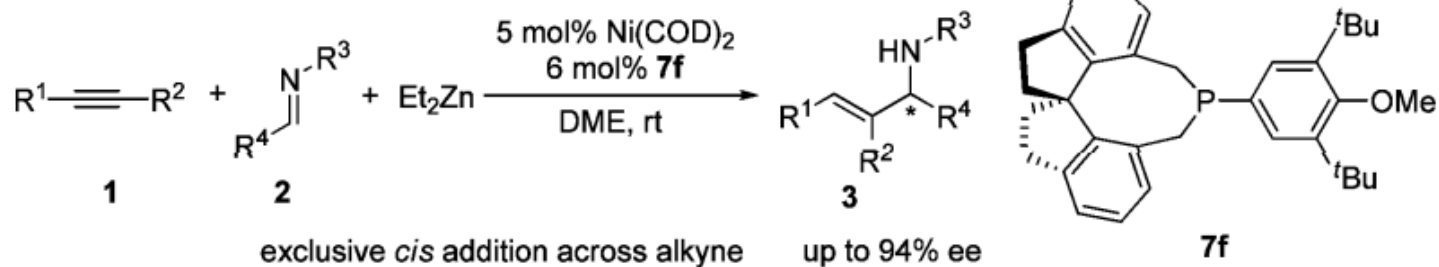


T. F. Jamison, *et al. Angew. Chem., Int. Ed.* **2003**, 42, 1364.

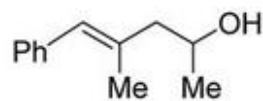
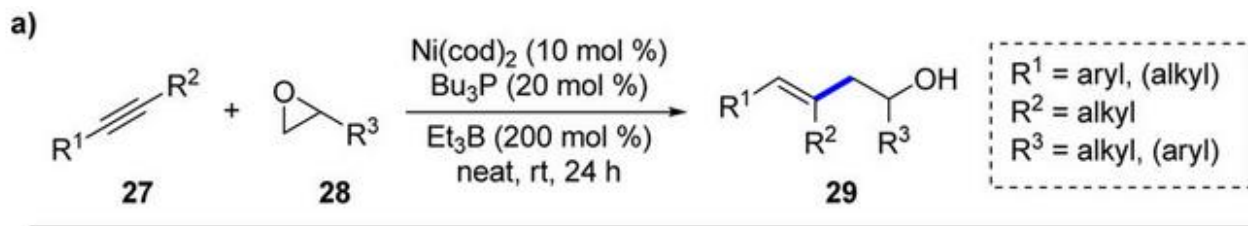
T. F. Jamison, *et al. Angew. Chem., Int. Ed.* **2004**, 43, 3941.



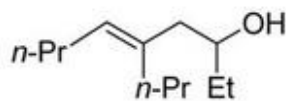
entry	R ³	[Ni]	ligand	solvent	time (h)	3/4/5 ^b	yield of 3 (%) ^c
1	Ts (2a)	Ni(acac) ₂	Ph ₃ P	THF	1	2:1:1	36
2	Ts (2a)	Ni(acac) ₂	(OPh) ₃ P	THF	1	2:1:10	5
3	Ts (2a)	Ni(acac) ₂	(<i>c</i> -C ₆ H ₁₁) ₃ P	THF	1	5:1:0.3	48
4	Ts (2a)	Ni(acac) ₂	<i>n</i> -Bu ₃ P	THF	1	23:1:3	66
5	Ts (2a)	Ni(acac) ₂	dppe ^d	THF	8	1:1:1	15
6	Ts (2a)	Ni(COD) ₂	<i>n</i> -Bu ₃ P	THF	1	24:1:3	86
7 ^e	Ts (2a)	Ni(COD) ₂	<i>n</i> -Bu ₃ P	DME	1	27:1:3	90
8 ^{ef}	Ts (2a)	Ni(COD) ₂	<i>n</i> -Bu ₃ P	DME	2	23:1:2	82
9 ^{ef}	 (2b)	Ni(COD) ₂	<i>n</i> -Bu ₃ P	DME	2	22:1:1	88
10 ^{ef}	 (2c)	Ni(COD) ₂	<i>n</i> -Bu ₃ P	DME	2	28:1:1	87



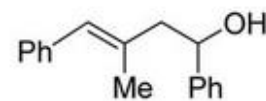
Reductive coupling of alkyne and epoxides:



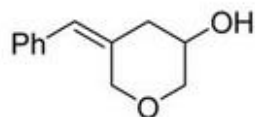
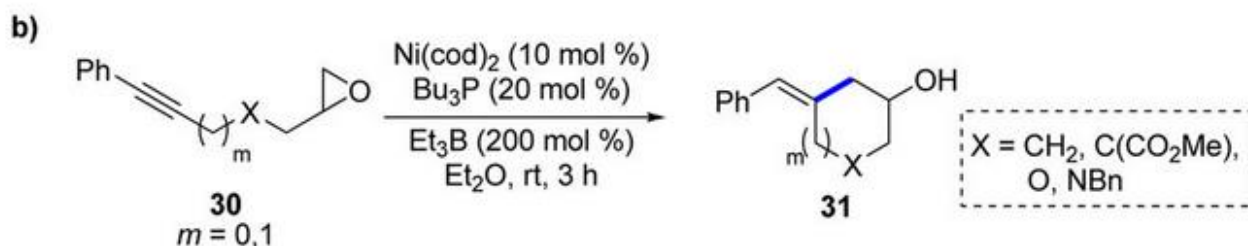
29a
71% yield
>95:5 rr (alkyne)
>95:5 rr (epoxide)



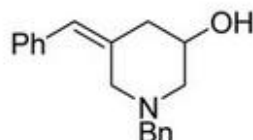
29b
35% yield
-- rr (alkyne)
>95:5 rr (epoxide)



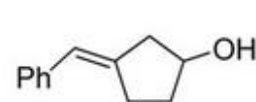
29c
50% yield
88:12 rr (alkyne)
83:17 rr (epoxide)



31a
50% yield
>95:5 rr (epoxide)



31b
65% yield
>95:5 rr (epoxide)

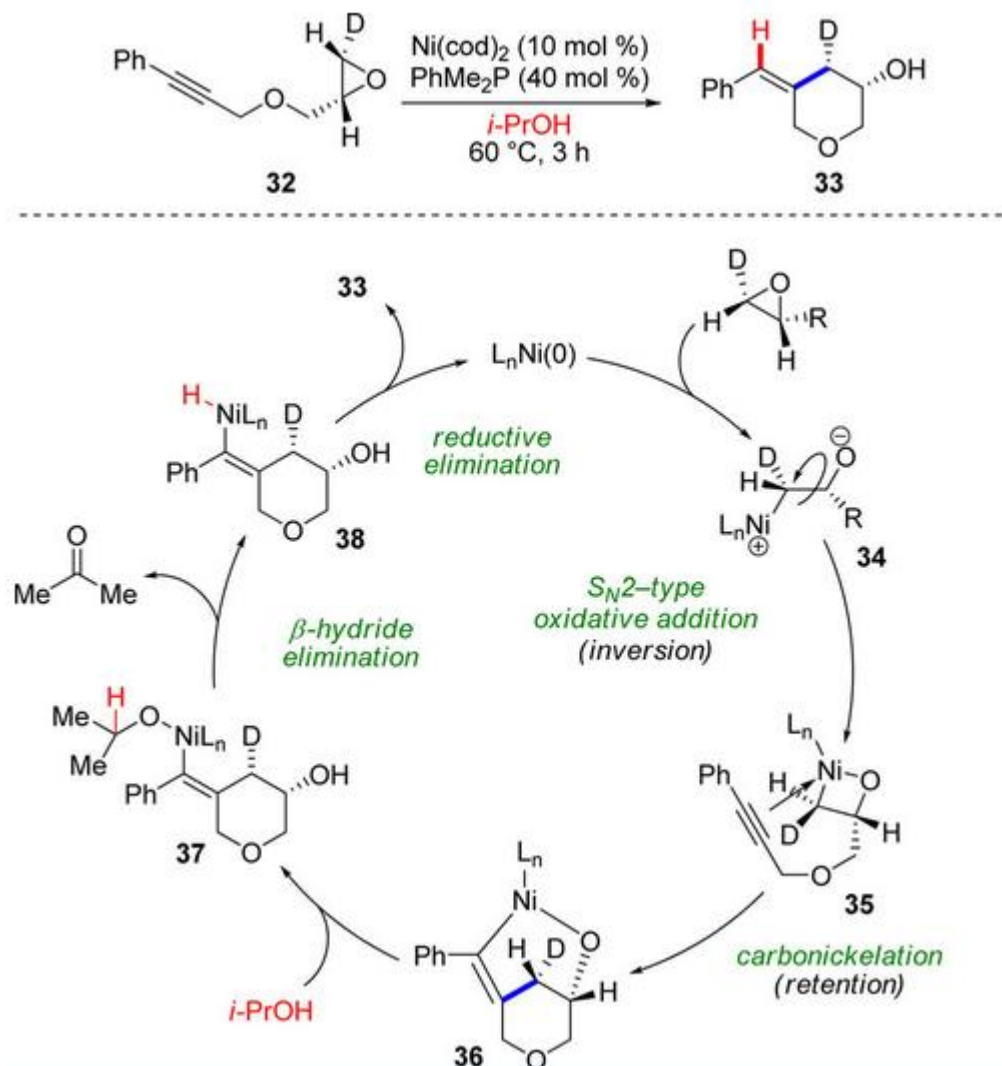


31c
50% yield
>95:5 rr (epoxide)

T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2003**, 125, 8076.

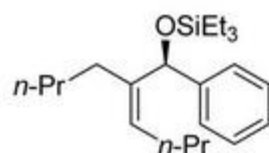
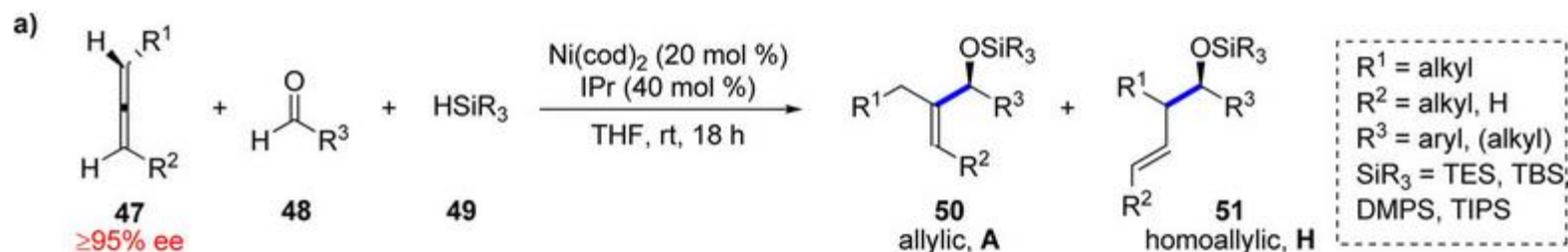
T. F. Jamison, *et al.* *Org. Lett.* **2011**, 13, 4140.

Reductive coupling of alkyne and epoxides: mechanism study

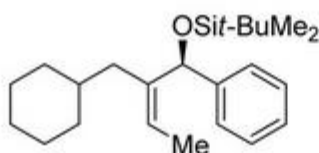


T. F. Jamison, *et al.* *Org. Lett.* **2011**, 13, 4140.

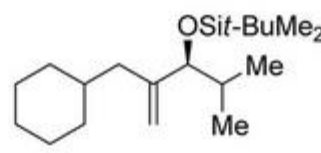
Reductive coupling of allene and aldehyde:



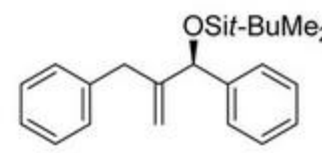
80% yield
94:6 A/H
>95:5 Z/E
95% ee



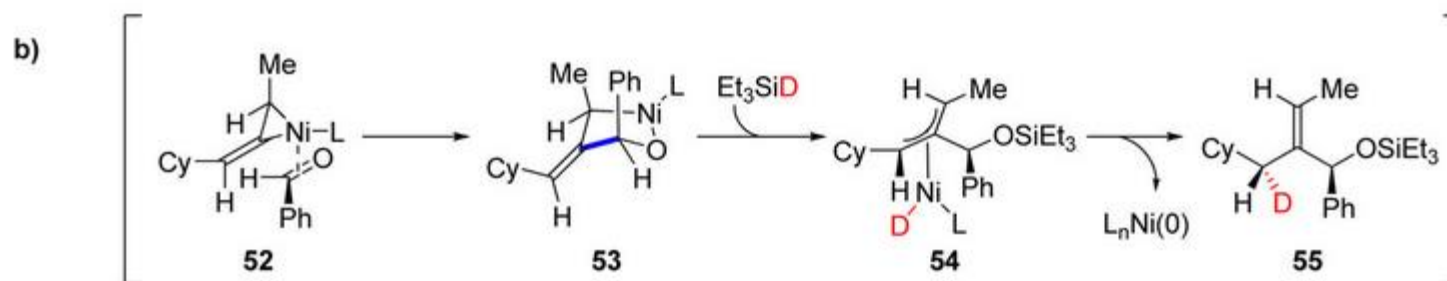
68% yield
90:10 A/H
>95:5 Z/E
98% ee



71% yield
>95:5 A/H
[Cyp₃P rather than IPr]

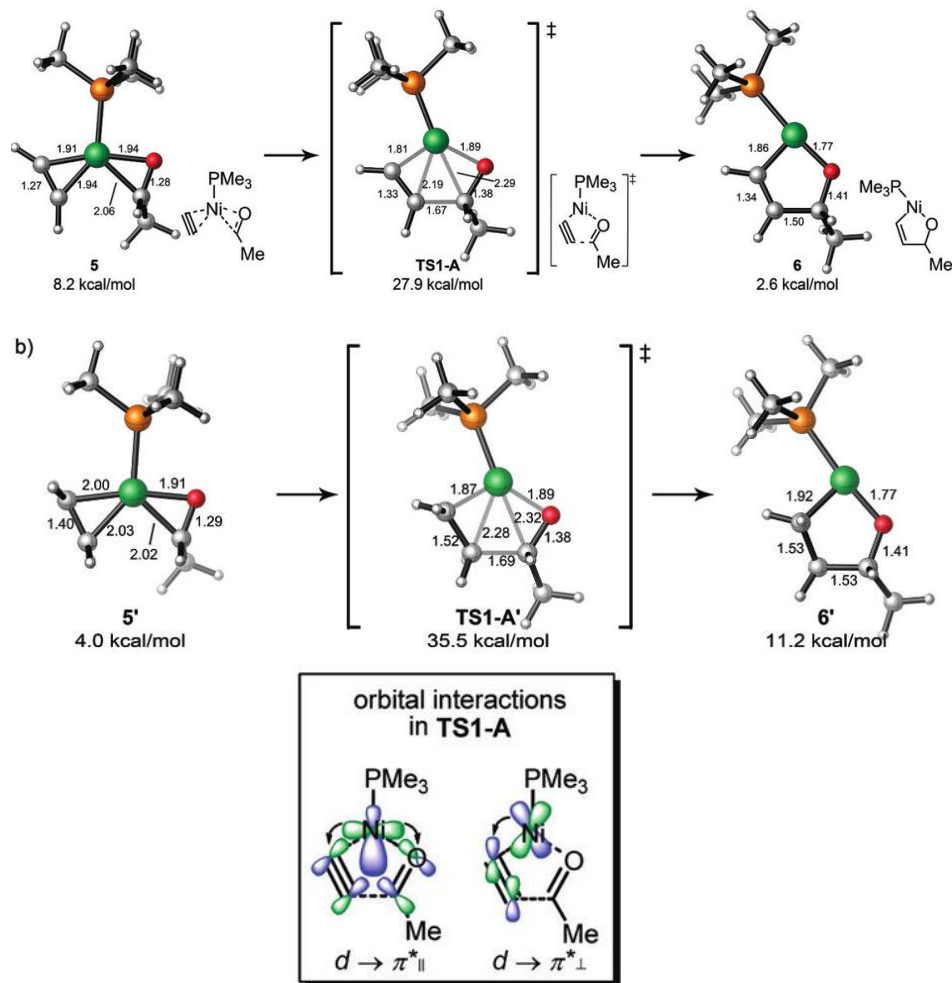


56% yield
>95:5 A/H

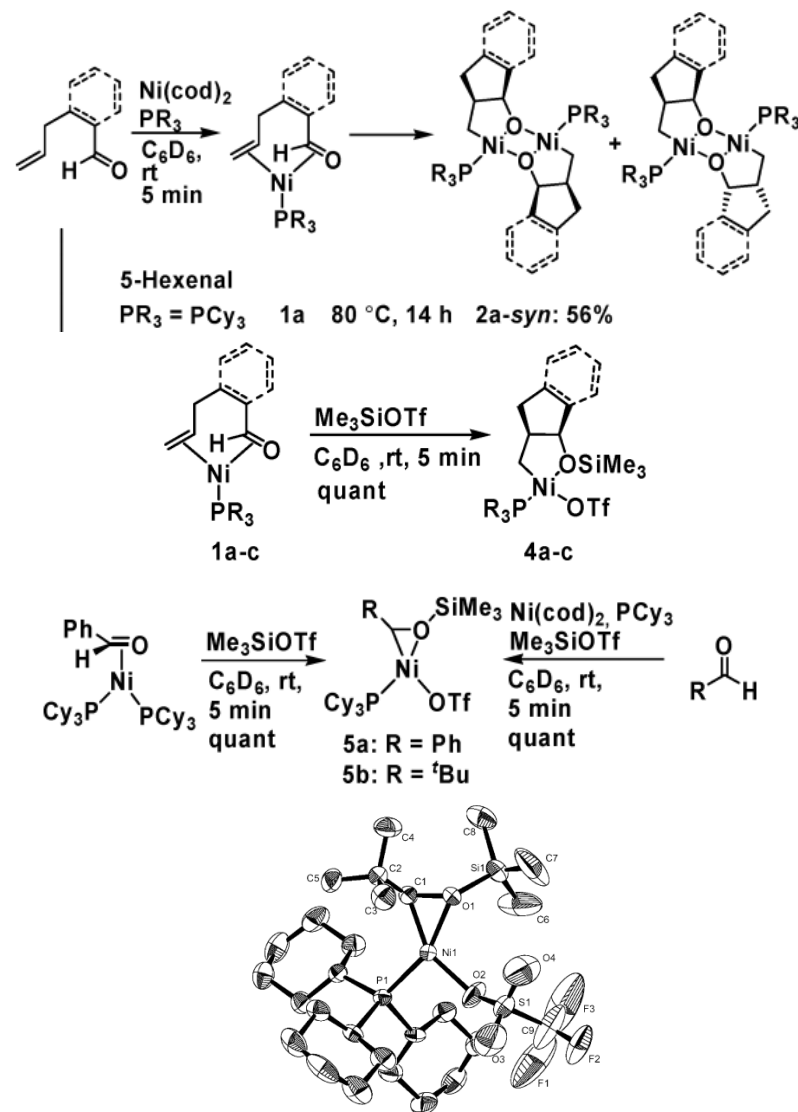


T. F. Jamison, *et al.* *J. Am. Chem. Soc.* **2005**, 127, 7320.

Reductive coupling of alkene and aldehyde:

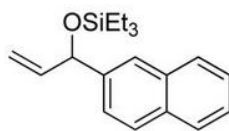
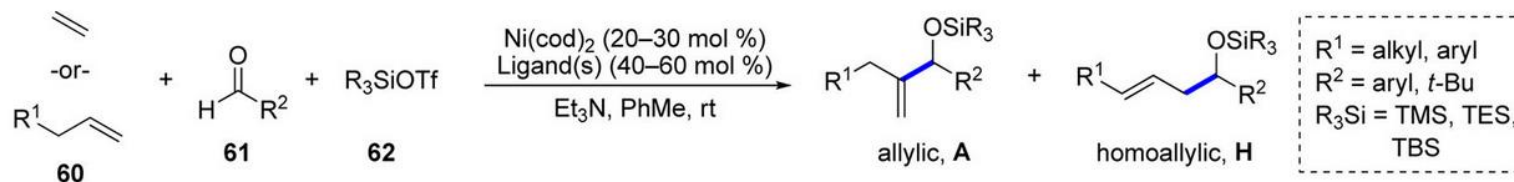


K. N. Houk and T. F. Jamison, *et al.*
J. Am. Chem. Soc. **2009**, 131, 6654.

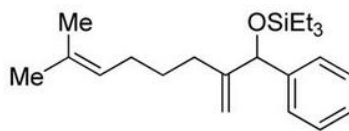


S. Ogoshi, *et al.* *J. Am. Chem. Soc.* **2004**, 126, 11802.
 S. Ogoshi, *et al.* *J. Am. Chem. Soc.* **2005**, 127, 12810.

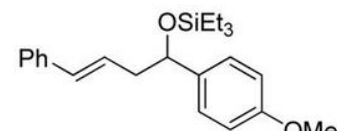
Reductive coupling of alkene and aldehyde:



63a
 $\text{L} = \text{P}(o\text{-anis})_3$
 95% yield



63b
 $\text{L} = \text{PCy}_2\text{Ph}$ 72% yield 69:31 A:H
 Moderate selectivity
 $\text{L} = \text{IPr} + \text{P(OPh)}_3$ 94% yield >95:5 A:H
 High selectivity



63c
 $\text{L} = \text{P}(\text{OEt})\text{Ph}_2$
 99% yield
 92:8 H:A, >95:5 E:Z

J. Am. Chem. Soc. **2005**, 127, 14194.

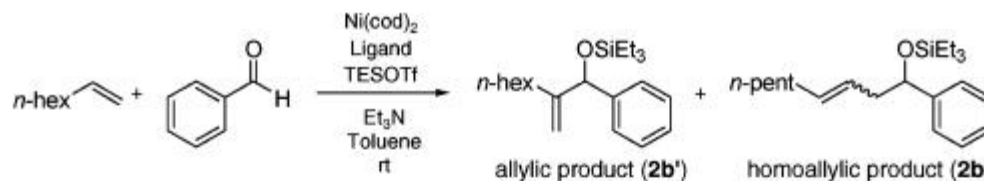
change ligand system

J. Am. Chem. Soc. **2006**, 128, 5362.

Angew. Chem., Int. Ed. **2007**, 46, 782.

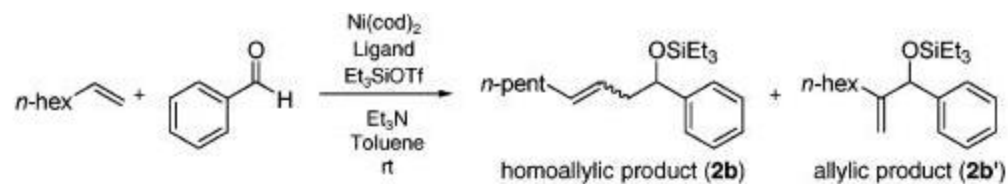
Conditions	Alkene	Ligand	Major Product
A	ethylene	$\text{P}(o\text{-anis})_3$	n/a
B	terminal alkene	PCy_2Ph	A
C	terminal alkene	PPh_3 or $\text{P}(\text{OEt})\text{Ph}_2$	H
D	terminal alkene	IPr and $\text{P}(\text{OPh})_3$	A

Reductive coupling of alkene and aldehyde: ligand effects



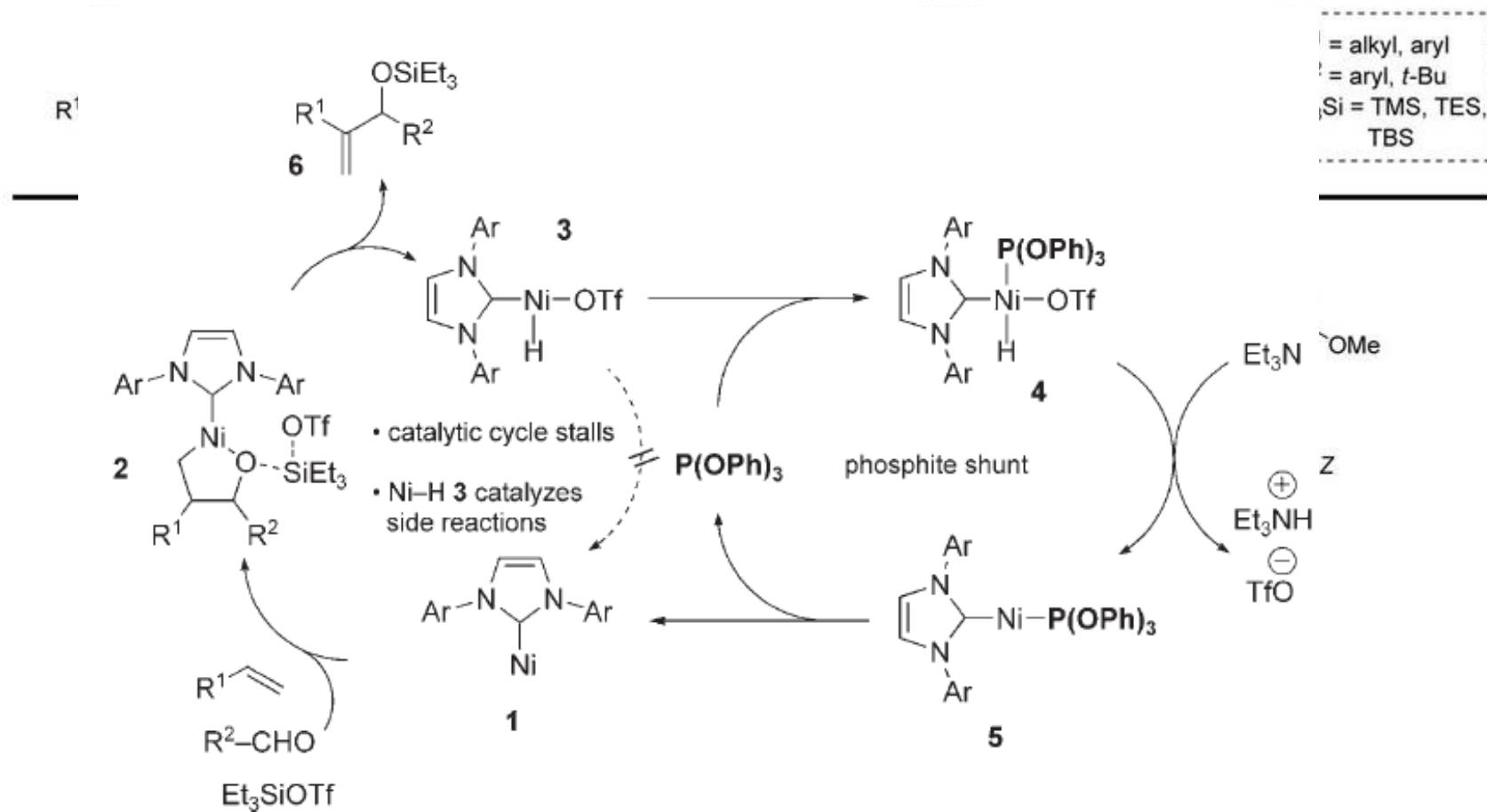
entry	ligand	cone angle ^b	ν_{CO} ^b	yield (2b') ^c	yield (2b) ^c	ratio (2b' : 2b) ^d	combined yield (2b' + 2b)
1	(<i>n</i> -Bu) ₃ P	132	1915	11%	27%	29:71	38%
2	(<i>n</i> -Oct) ₃ P	107		7%	14%	23:77	21%
3	(<i>i</i> -Pr) ₃ P	160	1915	11%	2%	85:15	13%
4	Cyp ₃ P			10%	3%	78:22	13%
5	Cy ₃ P	170	1915	13%	3%	81:19	16%
6	(<i>t</i> -Bu) ₃ P	182		7%	2%	78:22	9%
7	Cy ₂ PhP	162	1917	37%	16%	70:30	53%
8 ^e	Cy ₂ (<i>o</i> -tol)P	181		30%	5%	86:14	35%
9	Cy ₂ (<i>o</i> -Ph-Ph)P			32%	22%	60:40	54%
10 ^f	Cy ₂ FcP			14%	2%	88:12	16%

T. F. Jamison, *J. Am. Chem. Soc.* **2006**, 128, 11513.



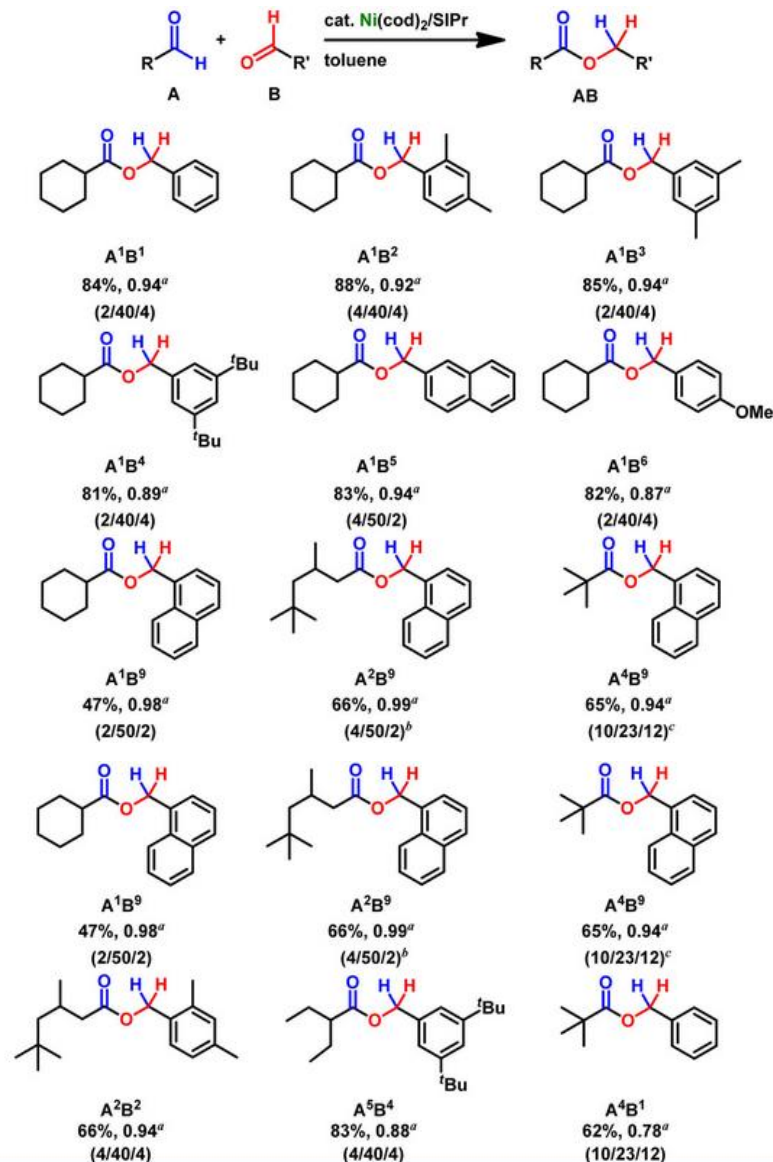
entry	ligand	cone angle ^b	ν_{CO} ^b	combined yield ^c (2b + 2b')	ratio (2b : 2b') ^d	<i>E</i> : <i>Z</i> (2b) ^d
1 ^e	Cy ₂ PhP	162	1917	73	29:71	n.d.
2	CyPh ₂ P	153	1917	84	75:25	91:9
3	(<i>o</i> -anisyl) ₃ P	194	1919	70	83:17	80:20
4	FcPh ₂ P	173		70	88:12	81:19
5	(<i>p</i> -tol) ₃ P	145	1920	78	92:8	67:33
6	Ph ₃ P	145	1922	73	92:8	67:33
7	(<i>p</i> -F-Ph) ₃ P	145	1924	74	92:8	57:43
8	(EtO)Ph ₂ P	133	1926	81	95:5	75:25
9	(<i>p</i> -CF ₃ -Ph) ₃ P	145	1929	44	>95:5	69:31
10	(EtO) ₂ PhP	121	1932	20	>95:5	89:11
11	(PhO) ₃ P	128	1951	<5	n.d.	n.d.

T. F. Jamison, *J. Am. Chem. Soc.* **2006**, *128*, 11513.



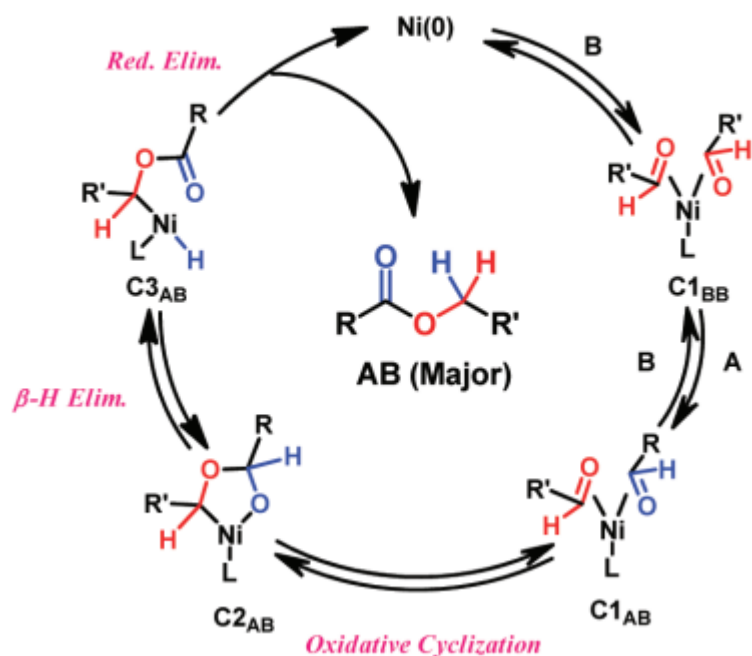
T. F. Jamison, *Angew. Chem., Int. Ed.* **2007**, 46, 782.

Crossed Tishchenko reaction:

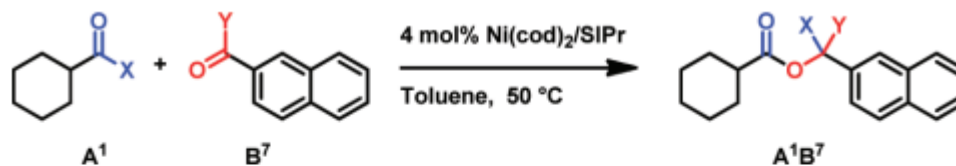
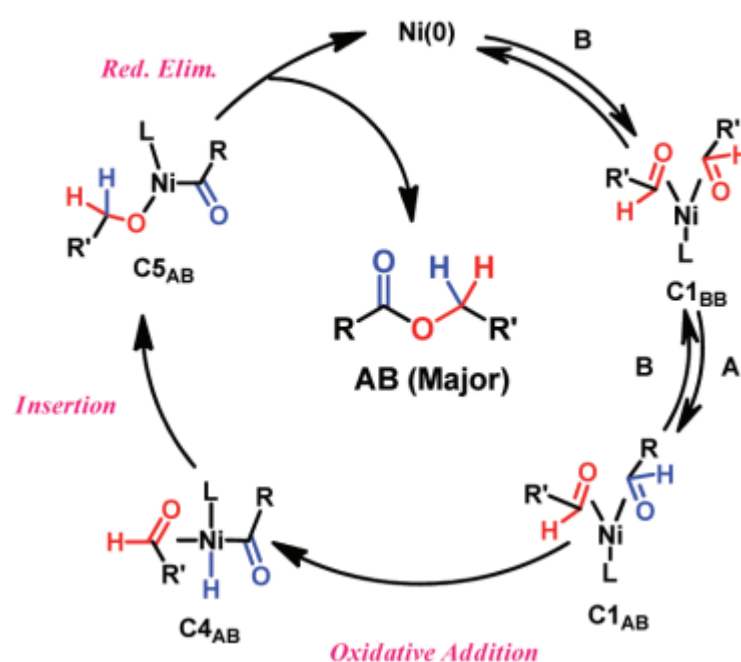


Crossed Tishchenko reaction: mechanism study

Path 2A

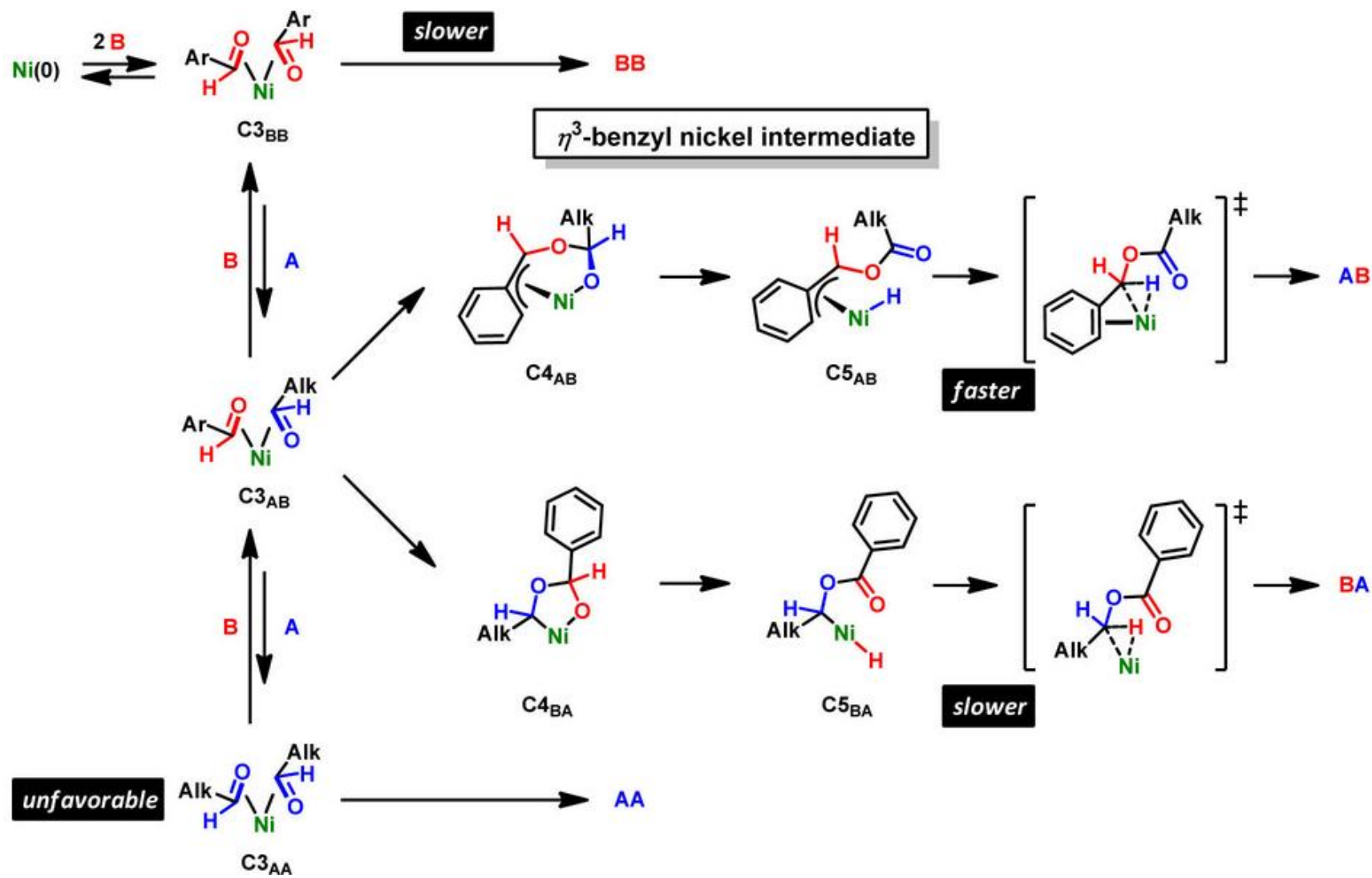


Path 2B



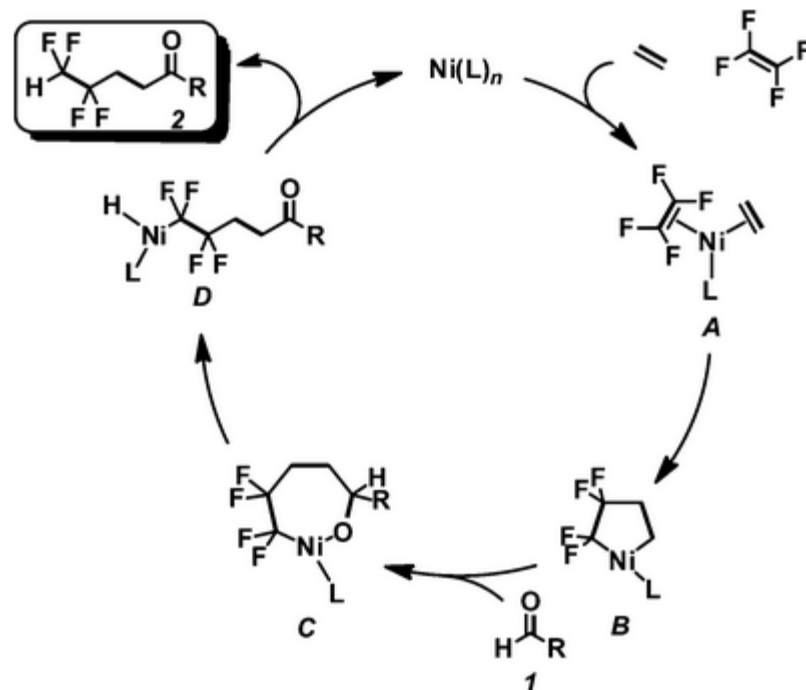
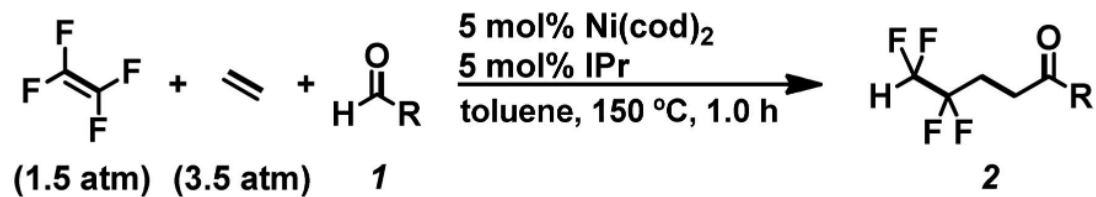
Experiment	X	Y	k_N^a	KIE(N) ^b	Selectivity
I	H	H	12.9(1)	—	0.94
II	H	D	10.9(1)	1.2	0.93
III	D	H	6.4(3)	2.0	0.83 ^c
IV	D	D	6.7(1)	1.9	0.85 ^d

S. Ogoshi, *et al.* *J. Am. Chem. Soc.* **2011**, 133, 4668.



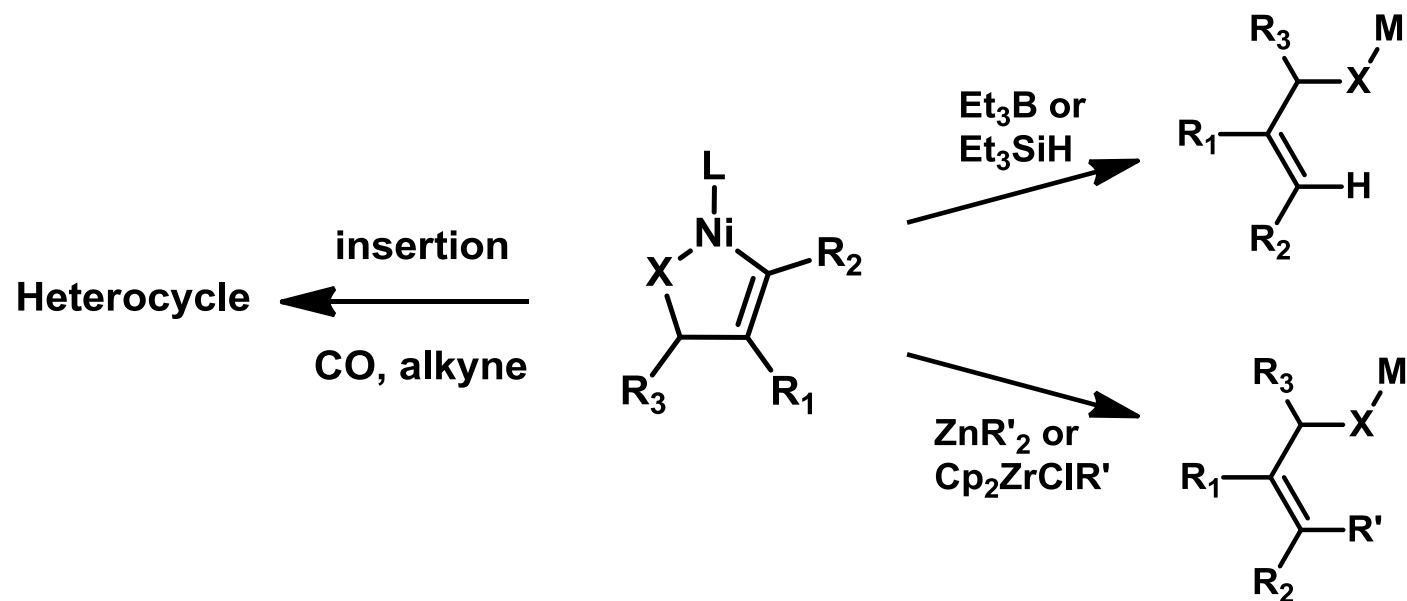
S. Ogoshi, *et al.* *J. Am. Chem. Soc.* **2011**, 133, 4668.

Cross-trimerization reaction of tetrafluoroethylene, ethylene, and aldehydes:

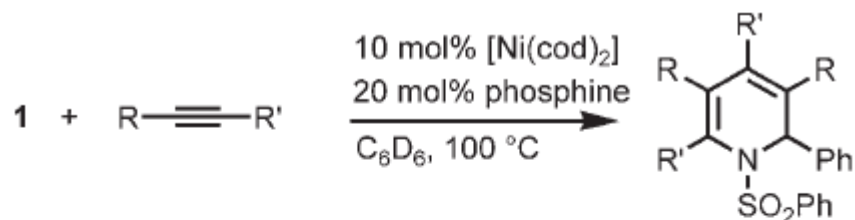
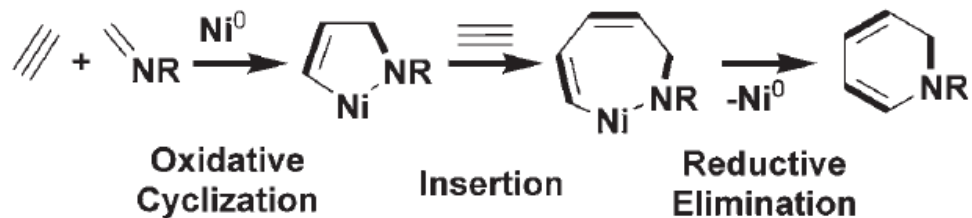


S. Ogoshi, *et al. J. Am. Chem. Soc.* **2015**, 137, ASAP.

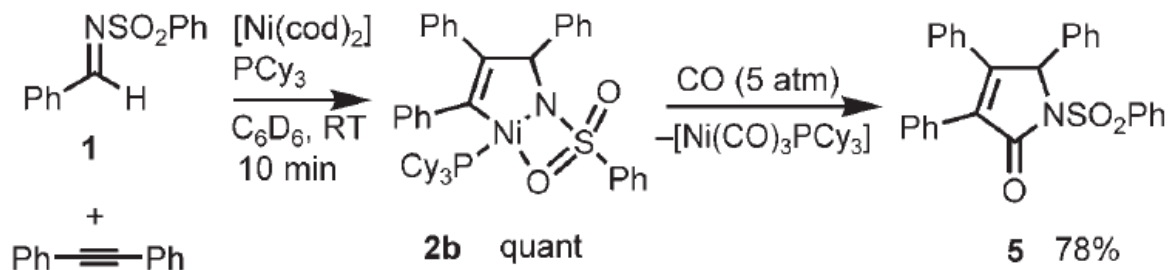
3. Cycloaddition Reaction



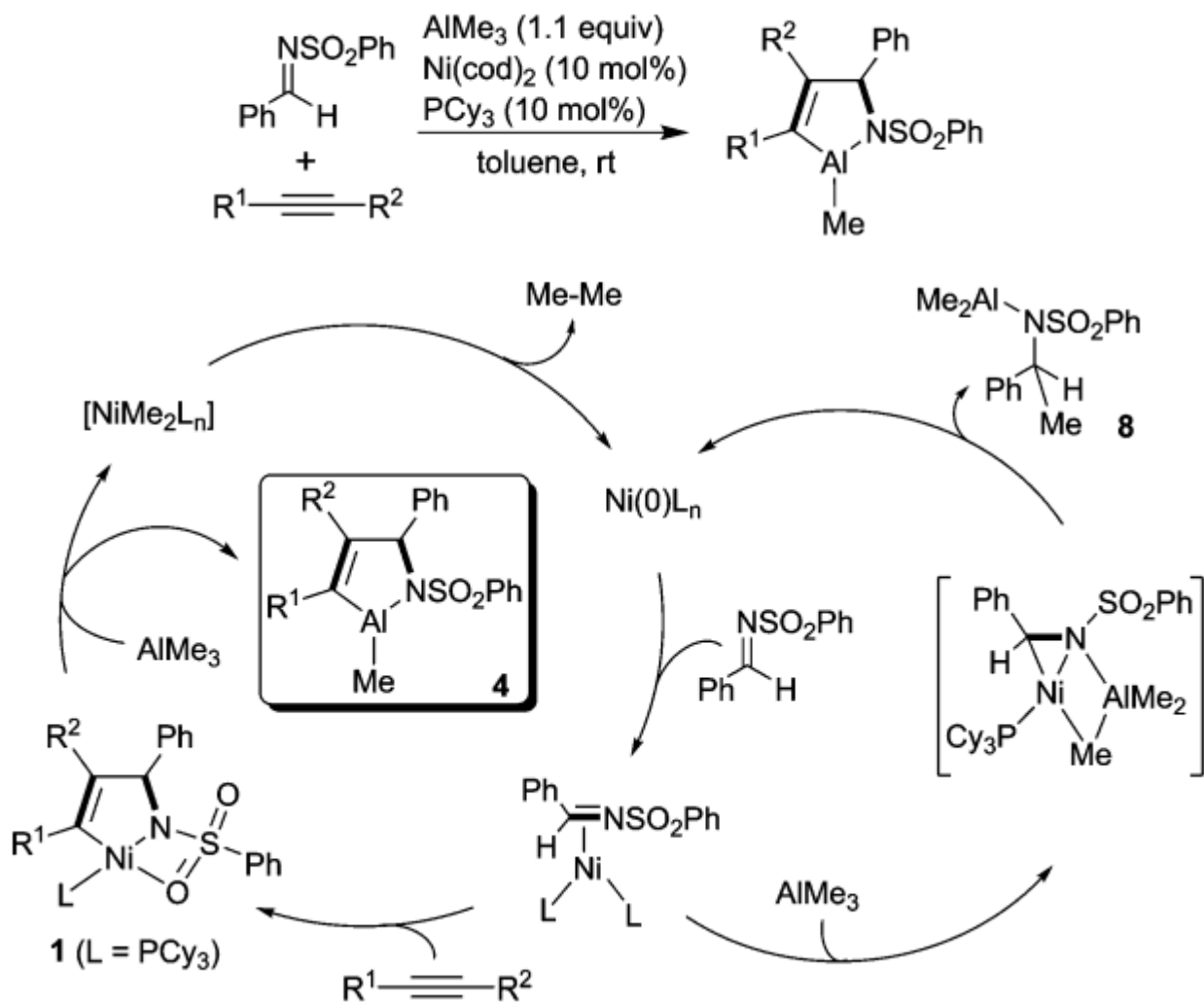
Formation of pyridine through 2+2+2:



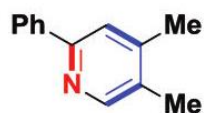
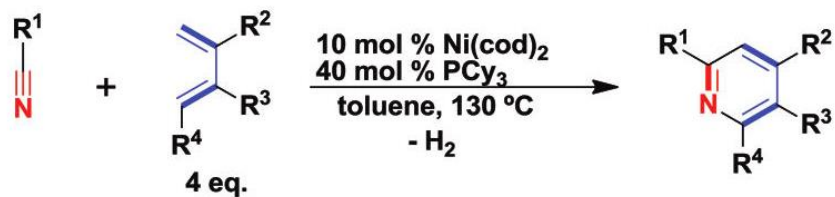
R, R' = Me	PMe _t Bu ₂	48 h	6a	87%
	PCy ₃	24 h		64%
	P ⁿ Bu ₃	24 h		17%
	P(o-tol) ₃	29 h		6%
R, R' = Et	PMe _t Bu ₂	70 h	6d	64%
	PCy ₃	24 h		26%
R = TMS, R' = H	PMe _t Bu ₂	18 h	6e	58%



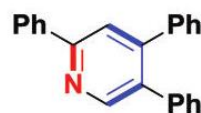
Formation of azaaluminacyclopentenes:



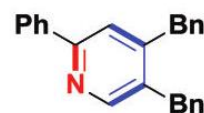
Formation of pyridine through 4+2:



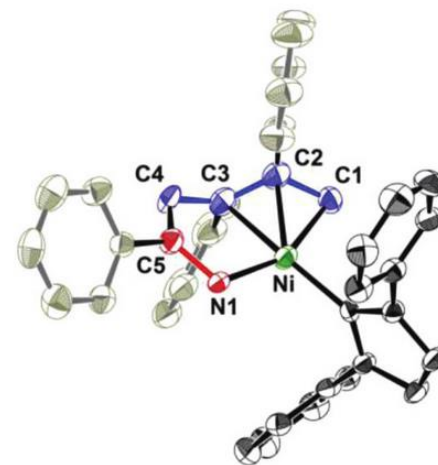
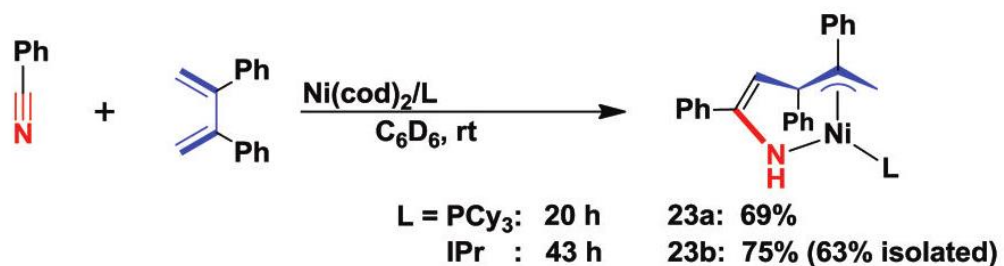
1 (88%; 48 h)



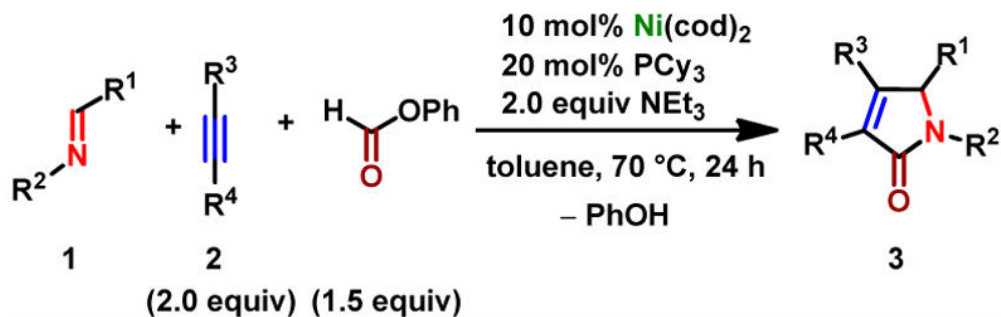
2 (81%; 12 h)



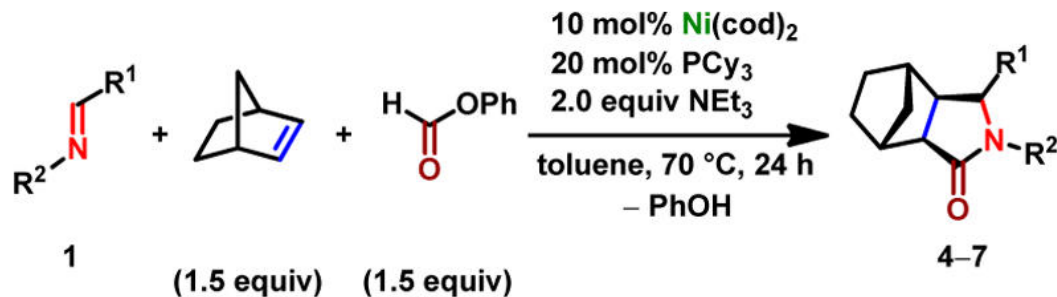
3 (77%; 38 h)^b



Formation of γ -lactams through 2+2+1:

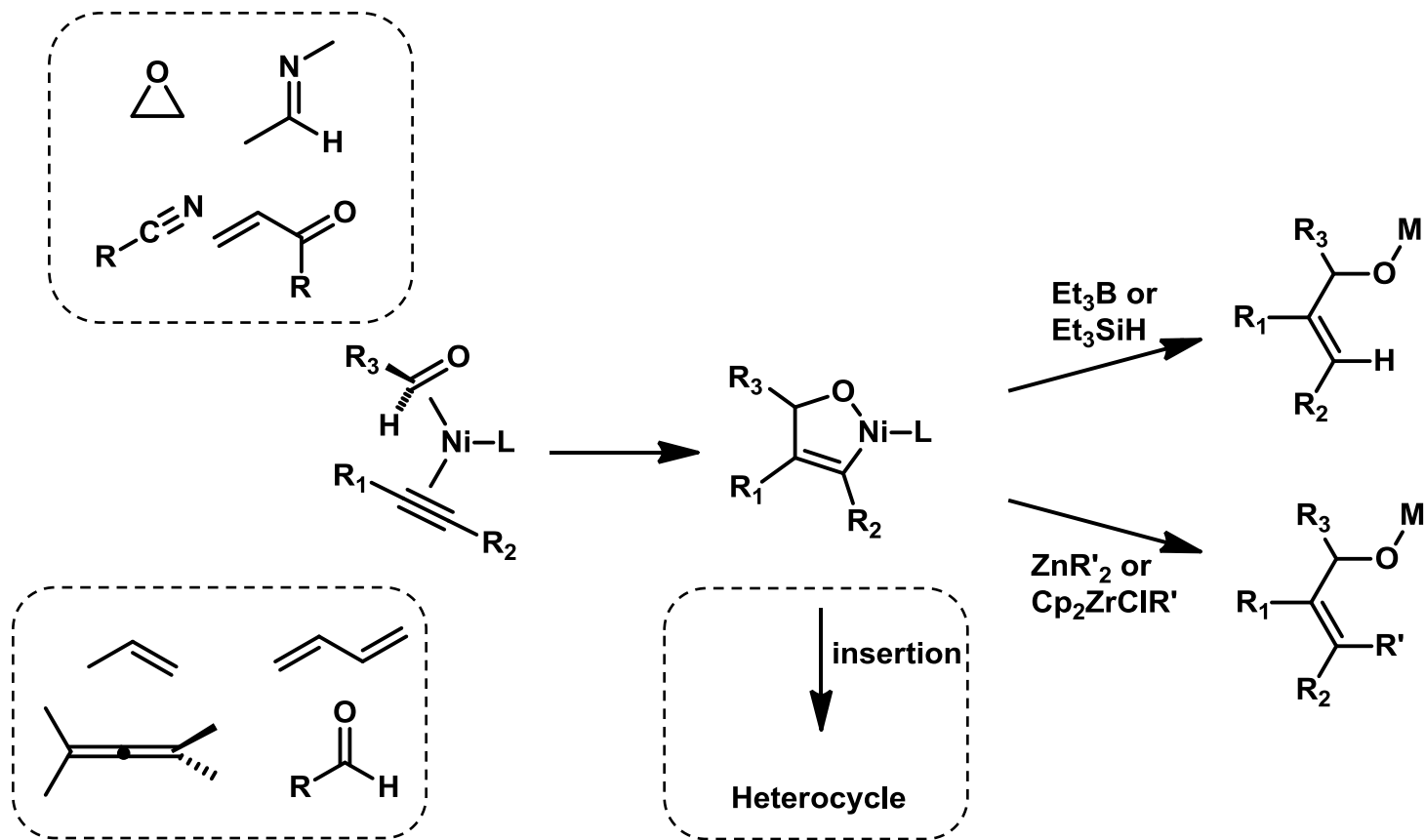


Scope of alkyne ($R^1 = Ph$, $R^2 = Bs = SO_2Ph$)



S. Ogoshi, *et al.* *J. Am. Chem. Soc.* **2014**, 136, 15877.

Summary



**Thanks for Your
Attention !**