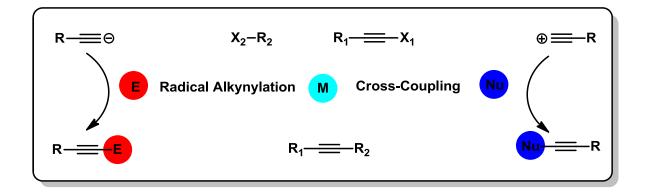
#### --New approaches to introduce an alkynyl group

--New approaches to introduce an alkynyl group



Reporter: Zhao-feng Wang Supervisor: Yong Huang 2013-03-27

#### Contents

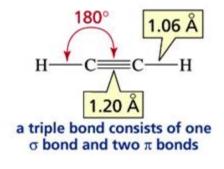
- 1. Introduction of Acetylene Chemistry
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#### Introduction -- Structure and Bonding

#### **Linear Acetylenic Scaffolds**

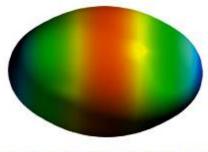
#### **p***K*<sub>a</sub> ≅ 25

Total bond strength: 839 kJ/mol C-C $\sigma$  bond: 369 kJ/mol 1<sup>st</sup> C-C $\pi$  bond: 268 kJ/mol 2<sup>nd</sup> C-C $\pi$  bond :202 kJ/mo



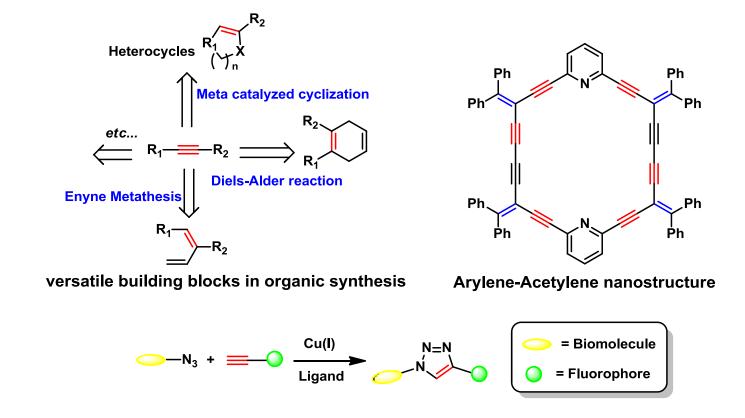


ball-and-stick model of ethyne



electrostatic potential map for ethyne

Introduction -- Why we need to introduce an alkynyl group?



click chemistry for biomolecular labeling

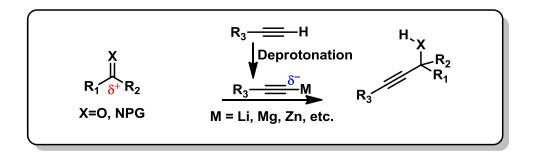
Tykwinsky, R. R. *et al. Synlett* **2004**, 182-184. Q. Lin *et al. Chem. Commun.* **2010**, *46*, 1589-1600

#### Contents

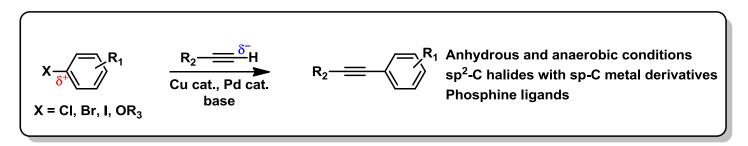
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Nucleophilic alkynylation: Classic text book approach

#### Addition of Alkyne Nucleophiles to Carbonyl Groups

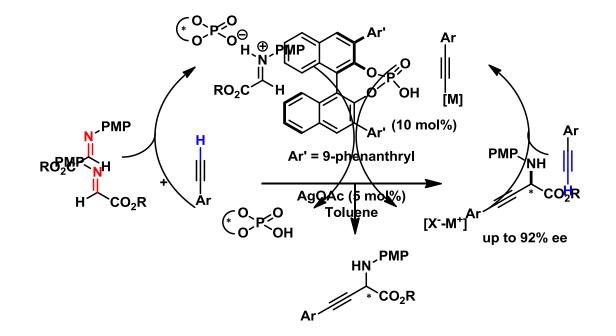


#### Sonogashira coupling of aryl halides and acetylenes



#### **Nucleophilic alkynylation**

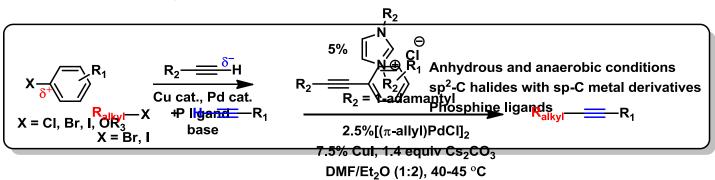
Asymmetric alkynylation of  $\alpha$ -imino esters via synergistic catalysis strategy



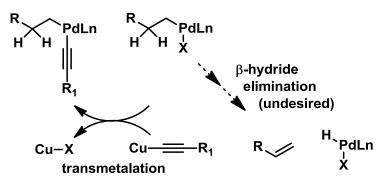
# Combined enantioselective Brønsted acid and metal-catalyzed alkynylation of $\alpha$ -imino esters

#### **Nucleophilic alkynylation**

Boroogtase iraprouplineg too fas phoglades and updet glenes



The first applications of carbene ligands in sonogashira reactions of unactivated alkyl halides

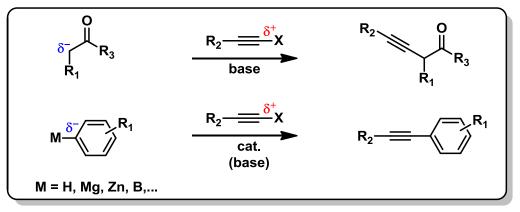


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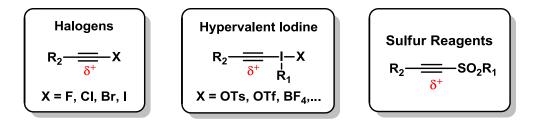
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#### Electrophilic alkynylation : The dark side of acetylene chemistry

Addition of alkynes on a nucleophilic position

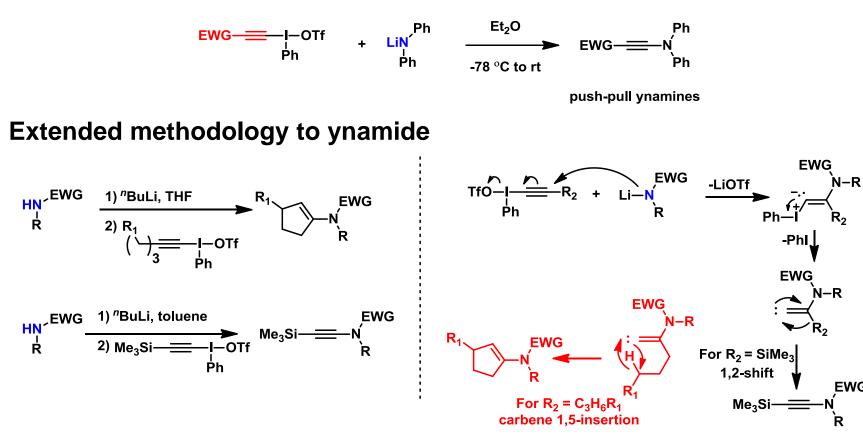


**Electrophilic alkynylation reagents** 



#### **Electrophilic alkynylation : Heteroatom alkynylation**

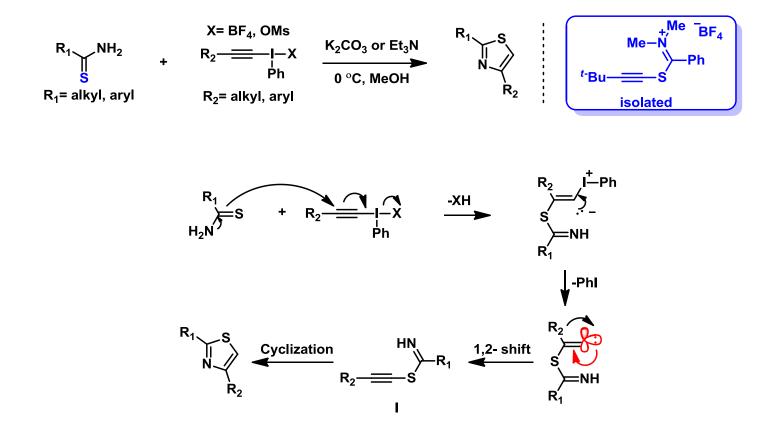
#### C-N bond formation: first ynamine synthesis



P. J. Stang *et al. Synthesis* 1994,1255–1256
K. S. Feldmen *et al. J. Org. Chem.*1996, *61*, 5440–5452
A. T. Stengel *et al. Angew. Chem., Int. Ed.* 1998, 37,489–492

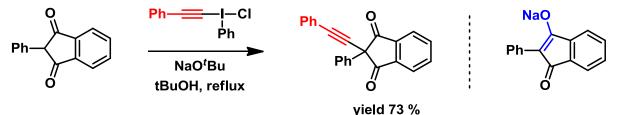
#### **Electrophilic alkynylation : Heteroatom alkynylation**

C-S bond formation: regiospecific thiazole synthesis using alkynyliodonium salts

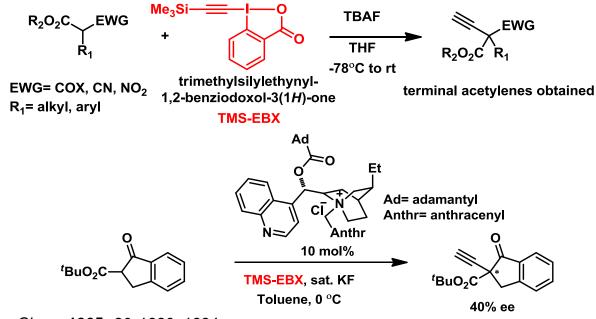


#### **Electrophilic alkynylation : Enolate alkynylation**

#### First alkynyliodonium salt reacted with the enolate



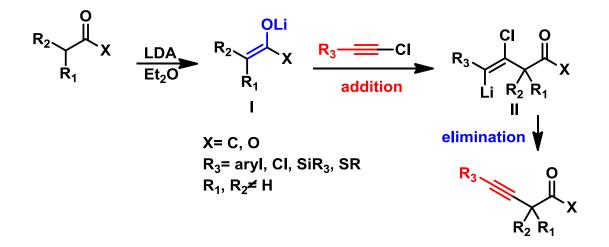
Improvement of the methodology involving novel hypervalent iodine reagents



S. A. Galton *et al. J. Org. Chem.* **1965**, *30*, 1930–1934 J. Waser *et al. Chem.–Eur. J.* **2010**, *16*, 9457–9461

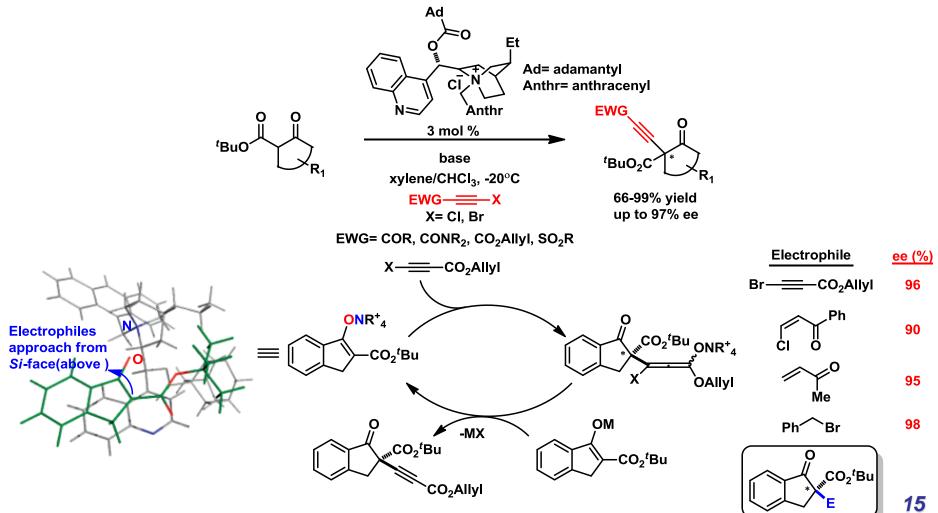
Electrophilic alkynylation : Enolate alkynylation

Alkynylation of non-stabilized enolates using chloroacetylenes



#### **Electrophilic alkynylation : Enolate alkynylation**

#### Highly enantioselective electrophilic alkynylation



K. A. Jørgensen et al. J. Am. Chem. Soc. 2007, 129, 441-449

**Electrophilic alkynylation : Alkynylation of organometallic nucleophiles** 

Pioneering work using alkynyl sulfones

$$R_1 \longrightarrow SO_2Ar + M-R_2 \xrightarrow{THF} R_1 \longrightarrow R_2$$
  
 $M = Li, MgX$   
 $R_1, R_2 = aryl, tertiary alkyl$ 

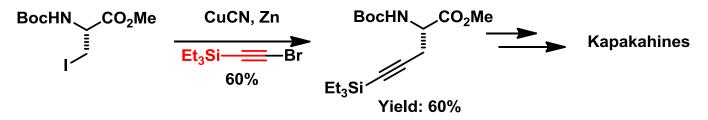
Efficient synthesis of aliphatic acetylenes based on a mixed Zn–Cu reagent

$$IZn-R_1 \xrightarrow{CuCN} ZnI(NC)Cu-R_1 \xrightarrow{R_2 - - - X} R_1 - - R_2$$

$$IZn-R_1 \xrightarrow{THF, -65^{\circ}C} R_1 - - R_2$$

$$X = Br, I$$

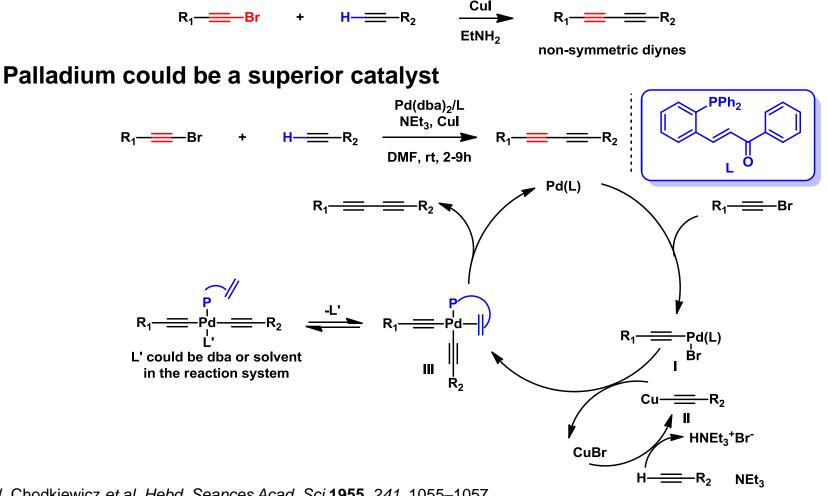
Organocopper reagent as nucleophile in total synthesis



W. E. Truce *et al. J. Org. Chem.* **1979**, *44*, 3444–3445
P. Knochel *et al. Tetrahedron Lett.* **1989**, *30*,4799–4802.
P. S. Baran *et al. J. Am.Chem. Soc.* **2010**, *132*, 7119–7137.

Electrophilic alkynylation : Alkynylation of C(sp)-H

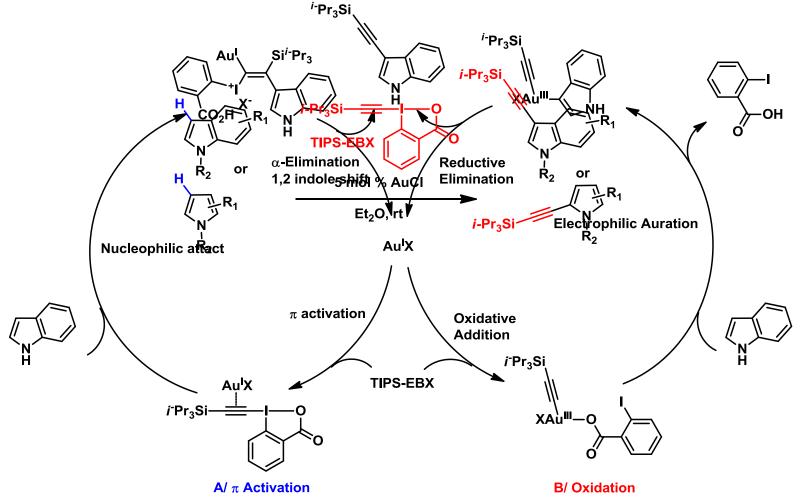
Cadiot–Chodkiewicz alkynylation of terminal alkynes



W. Chodkiewicz *et al. Hebd. Seances Acad. Sci.***1955**, *241*, 1055–1057. A.-W. Lei *et al. J. Am. Chem. Soc.* **2008**, *130*, 14713–14720

#### Electrophilic alkynylation : Alkynylation of C(sp<sup>2</sup>)–H

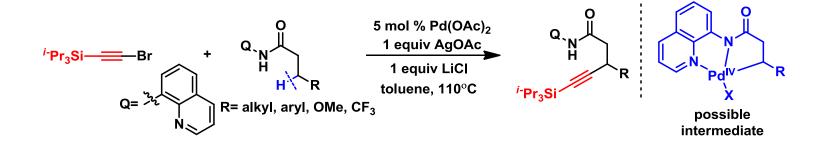
Gold-catalyzed alkynylation of indoles and pyrroles using alkynyl benziodoxolone



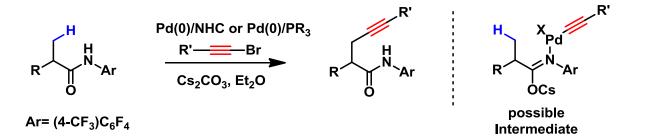
J. Waser et al. Angew. Chem., Int. Ed. 2009, 48, 9346–9349

Electrophilic alkynylation : Alkynylation of C(sp<sup>3</sup>)–H

First Palladium(II)-catalyzed  $\beta$ -C(sp<sup>3</sup>)–H bond alkynylation



Palladium(0)-catalyzed primary  $\beta$ -C(sp3)–H bond alkynylation



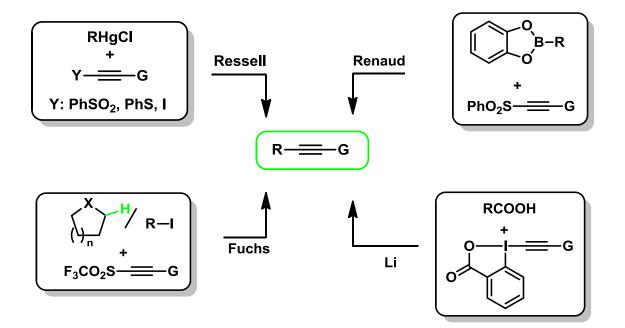
N. Chatani *et al. J. Am. Chem. Soc.* **2011**, *133*, 12984–12986 J.-Q. Yu *et al. J. Am. Chem. Soc.* **2013**, *135*, 3387–3390

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#### **Radical C-alkynylation**

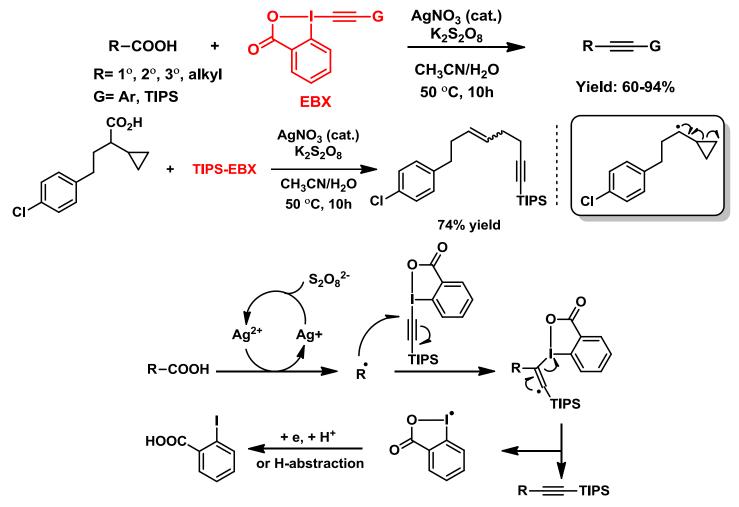
**Overview of radical C-alkynylation reactions** 



G. A. Russell *et al. Tetrahedron Lett.* **1986**, *27*, 3479-3482 P. L. Fuchs *et al. J. Am. Chem. Soc.* **1996**, *118*, 4486-4487 P. Renaud *et al. Angew. Chem., Int. Ed.* **2006**, *45*, 5847-5849 C.-Z. Li *et al. J. Am. Chem. Soc.* **2012**, *134*, 14330–14333

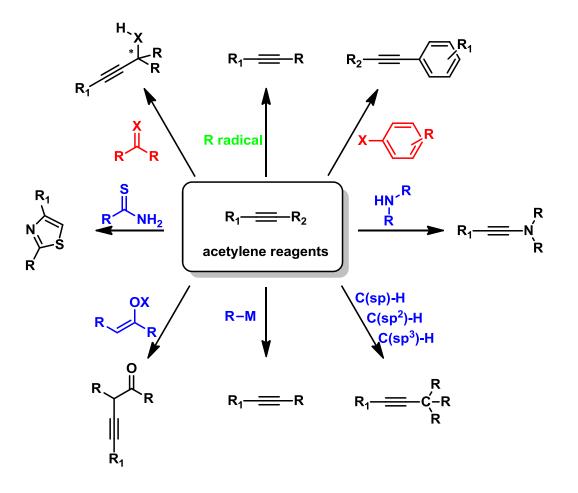
#### **Radical C-alkynylation**

#### Radical-mediated C(sp<sup>3</sup>) - C(sp) coupling



C.-Z. Li et al. J. Am. Chem. Soc. 2012, 134, 14330-14333

Summary



#### Acknowledgement



Prof. Yong Huang



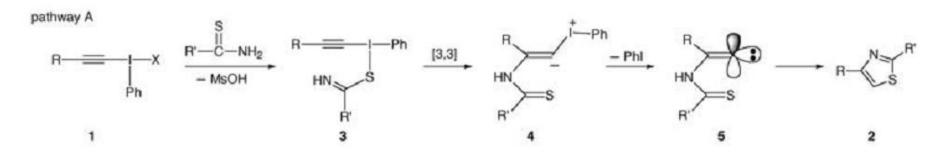
All my labmates in E201

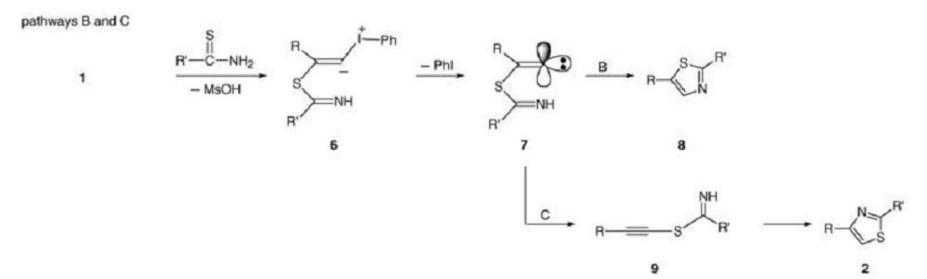


All the members in SCBB



## Supporting information

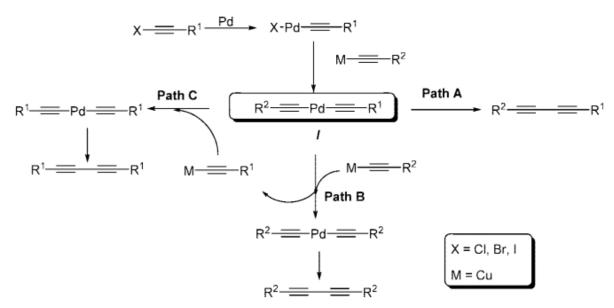




Scheme 2. Possible reaction mechanisms.

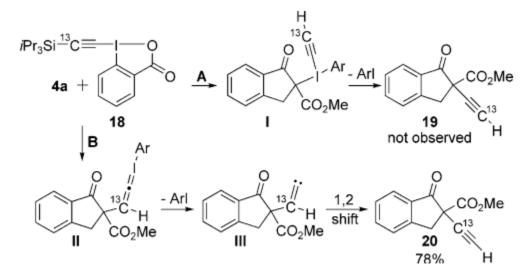


Scheme 2. Proposed Pathways of Palladium-Catalyzed C(sp)-C(sp) Coupling

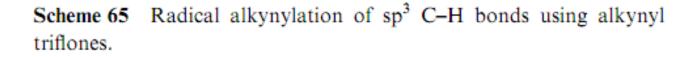


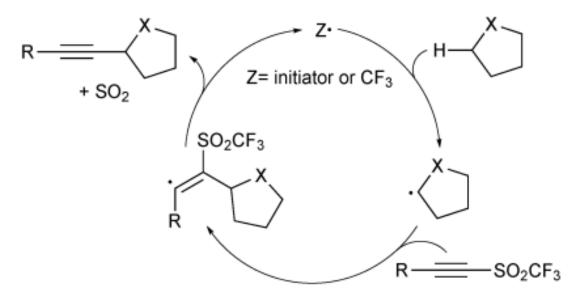
## Supporting information

General Procedures for the Coupling Reactions. To an ovendried Schlenk tube with a magnetic stir bar were added Pd(dba)2 (11.5 mg, 0.02 mmol), L1 ligand (7.9 mg, 0.02 mmol), and Cul (1.9 mg, 0.01 mmol). DMF (1 mL) was added via a syringe. The system was vacuumed with an oil pump at 0  $^{\circ}$  C and filled with nitrogen, and this was repeated five times. After the mixture was stirred under nitrogen for about 10 min, alkyne (0.6 mmol) was added via a microliter and stirred for another 5 min. 1-Bromoalkyne was added last via a microliter syringe. The system was stirred at room temperature for 10 h. Upon completion, 4 mL of brine was added, and the mixture was extracted by ethyl acetate (3 mL  $\times$  3). The product was obtained by flash column chromatography.



Scheme 3. Possible mechanisms for the ethynylation reaction and labeling experiment (Ar = phenyl-2-carboxylate).





Scheme 66 Mechanism of the radical alkynylation.